



**PHYSICAL SCIENCES (CHEMISTRY)  
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

**TIME: 1 HOUR**

**VERIFICATION OF BOYLE'S LAW  
INDIVIDUAL WORK (should be done under strict supervision).**

**1. Write a scientific report which must include the following:**

**1.1 Aim**

**(1)**

**1.2 Apparatus and method**

**(5)**

**2. Observations and Recording of results** (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate  $pV$  for each set of results and record your answers in the table.

Volume (V) $\text{cm}^3$	Pressure (p) kPa		$pV$ (J)

**(6)**

**3. Analysis of the results:**

**3.1 Identify the following variables for the experiment:**

- Independent
- Dependent
- Controlled

**(3)**

**3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus  $\frac{1}{p}$ .**

**(18)**

**4. Conclusion**

**4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.**

**(1)**

**4.2 Write down a conclusion for this experiment.**

**(3)**

**5**

**Questions and discussion:**

**5.1 Explain why you have to wait for a few minutes before you can take readings.**

**(2)**

**5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.**

**(4)**

# Physical Science Grade 11 2013 November Question Paper Capricorn

**Ying Liu**



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## **Table of Contents Physical Science Grade 11 2013 November Question Paper Capricorn**

1. Understanding the eBook Physical Science Grade 11 2013 November Question Paper Capricorn
  - The Rise of Digital Reading Physical Science Grade 11 2013 November Question Paper Capricorn
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade 11 2013 November Question Paper Capricorn
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Grade 11 2013 November Question Paper Capricorn
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade 11 2013 November Question Paper Capricorn
  - Personalized Recommendations
  - Physical Science Grade 11 2013 November Question Paper Capricorn User Reviews and Ratings
  - Physical Science Grade 11 2013 November Question Paper Capricorn and Bestseller Lists
5. Accessing Physical Science Grade 11 2013 November Question Paper Capricorn Free and Paid eBooks
  - Physical Science Grade 11 2013 November Question Paper Capricorn Public Domain eBooks
  - Physical Science Grade 11 2013 November Question Paper Capricorn eBook Subscription Services
  - Physical Science Grade 11 2013 November Question Paper Capricorn Budget-Friendly Options

6. Navigating Physical Science Grade 11 2013 November Question Paper Capricorn eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science Grade 11 2013 November Question Paper Capricorn Compatibility with Devices
  - Physical Science Grade 11 2013 November Question Paper Capricorn Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Grade 11 2013 November Question Paper Capricorn
  - Highlighting and Note-Taking Physical Science Grade 11 2013 November Question Paper Capricorn
  - Interactive Elements Physical Science Grade 11 2013 November Question Paper Capricorn
8. Staying Engaged with Physical Science Grade 11 2013 November Question Paper Capricorn
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Grade 11 2013 November Question Paper Capricorn
9. Balancing eBooks and Physical Books Physical Science Grade 11 2013 November Question Paper Capricorn
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Grade 11 2013 November Question Paper Capricorn
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Grade 11 2013 November Question Paper Capricorn
  - Setting Reading Goals Physical Science Grade 11 2013 November Question Paper Capricorn
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Grade 11 2013 November Question Paper Capricorn
  - Fact-Checking eBook Content of Physical Science Grade 11 2013 November Question Paper Capricorn
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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