

# Physical Science Curriculum

## Unit Sequence

SEMESTER ONE	SEMESTER TWO
<p><b>Unit 1: Scientific Method</b></p> <ul style="list-style-type: none"><li>• Lab Basics: Safety and equipment</li><li>• Measurement</li><li>• Dimensional analysis and scientific notation</li><li>• Scientific method</li></ul> <p><b>Unit 2: Motion and Force</b></p> <ul style="list-style-type: none"><li>• Describing motion</li><li>• Acceleration</li><li>• Newton's laws</li></ul> <p><b>Unit 3: Energy</b></p> <ul style="list-style-type: none"><li>• Nature of energy</li><li>• Conservation of energy</li><li>• Thermal energy</li><li>• Work and power</li></ul> <p><b>Unit 4: Electricity and Magnetism</b></p> <ul style="list-style-type: none"><li>• Charge</li><li>• Current</li><li>• Circuits</li><li>• Magnets</li></ul> <p><b>Unit 5: Waves</b></p> <ul style="list-style-type: none"><li>• Nature and properties of waves</li><li>• Behaviors of waves</li><li>• Electromagnetic spectrum</li></ul>	<p><b>Unit 6: Matter</b></p> <ul style="list-style-type: none"><li>• Classification of matter</li><li>• Properties of matter</li><li>• States of matter</li><li>• Solutions</li></ul> <p><b>Unit 7: Atomic Structure</b></p> <ul style="list-style-type: none"><li>• Structure of the atom</li><li>• Isotopes</li><li>• Periodic table of elements</li></ul> <p><b>Unit 8: Bonding</b></p> <ul style="list-style-type: none"><li>• Stability and bonding</li><li>• Naming ionic compounds</li><li>• Naming covalent compounds</li></ul> <p><b>Unit 9: Reactions</b></p> <ul style="list-style-type: none"><li>• Balancing reactions</li><li>• Classifying reactions</li><li>• Energy in reactions</li><li>• Acids and bases <i>(if time)</i></li><li>• Nuclear chemistry <i>(if time)</i></li></ul>

# Physical Science Final Exam Scope Grade 1

**TD Snyder**



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