## POGIL: Interaction of Radiation and Matter

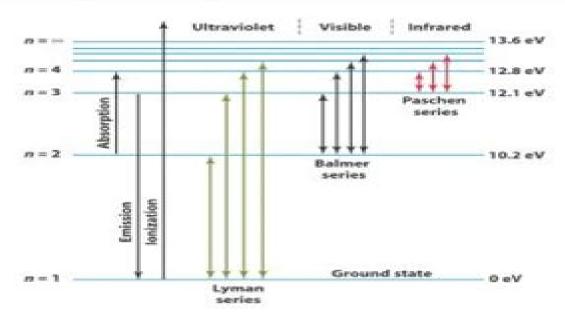
#### Why?

Light, which is also called electromagnetic radiation, carries energy and has characteristics of both particles and waves. The absorption and emission of electromagnetic radiation is one of the most powerful tools used to probe melecular structure and chemical reactions. It forms the basis of medicine's nuclear magnetic resonance imaging and is intrinsic to many analytical techniques used to monitor manufacturing and the environment. You will use the emission and absorption of light by matter to obtain the relative energies of the atomic orbitals and other energy levels that are associated with the vibration and rotation of molecules, and with electron and nuclear spin.

#### Model 1: Electronic Transitions in the Hydrogen Atom

The diagram below illustrates some possible transitions of the electron in a hydrogen atom.

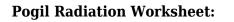
- In the Lyman series, electrons transition between higher energy levels and the first energy level resulting in ultraviolet radiation.
- In the Balmer series, electrons transition between higher energy levels and the second energy level resulting in visible light.
- In the Paschen series, electrons transition between higher energy levels and the third energy level resulting in infrared radiation.



Give complete answers to the key questions, exercises and problems. Provide explanations and show work, where necessary.

# **Pogil Radiation Worksheet**

**Michael Hoelscher** 



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