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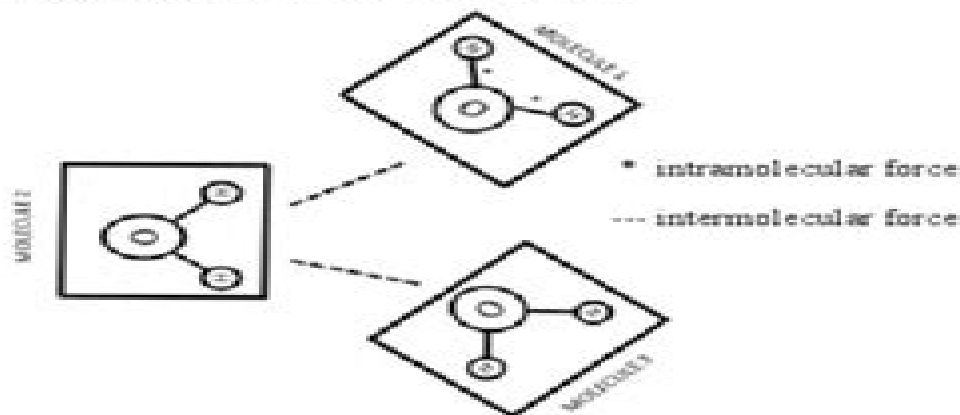
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POGIL: Intermolecular Forces

Model 1: What is an intermolecular force?

As you have learned, matter is made up of discrete particles called **atoms**, which chemically combine to form **molecules**. Molecules do not exist as independent units; in fact, groups of molecules "stick together" in order to form liquids and solids. The forces that hold groups of molecules together are **intermolecular forces**. Without intermolecular forces, the world as we know it would not be the same.

Figure 1: Intramolecular and Intermolecular Forces



Critical Thinking Questions:

1. What specific molecule is represented inside each box in Figure 1?
2. In relation to the box for molecule 1, where do the intramolecular forces exist in Figure 1 – inside the boxes or outside the boxes?
3. Based on the intramolecular forces for molecule 1, draw similar asterisks (*) for the intramolecular forces on the diagram for molecules 2 and 3.
4. In relation to the molecule, where do intramolecular forces tend to occur – within the molecule or outside of the molecule?
5. Two intermolecular forces exist in Figure 1. Where are they positioned relative to the molecules – within the molecules or between the molecules?
6. State the difference between intermolecular and intramolecular forces in terms of where they occur on the molecular level.



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Table of Contents Pogil Packet Answers Inheritance

1. Understanding the eBook Pogil Packet Answers Inheritance
 - The Rise of Digital Reading Pogil Packet Answers Inheritance
 - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Packet Answers Inheritance
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pogil Packet Answers Inheritance
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Packet Answers Inheritance

- Personalized Recommendations
- Pogil Packet Answers Inheritance User Reviews and Ratings
- Pogil Packet Answers Inheritance and Bestseller Lists
- 5. Accessing Pogil Packet Answers Inheritance Free and Paid eBooks
 - Pogil Packet Answers Inheritance Public Domain eBooks
 - Pogil Packet Answers Inheritance eBook Subscription Services
 - Pogil Packet Answers Inheritance Budget-Friendly Options
- 6. Navigating Pogil Packet Answers Inheritance eBook Formats
 - ePub, PDF, MOBI, and More
 - Pogil Packet Answers Inheritance Compatibility with Devices
 - Pogil Packet Answers Inheritance Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pogil Packet Answers Inheritance
 - Highlighting and Note-Taking Pogil Packet Answers Inheritance
 - Interactive Elements Pogil Packet Answers Inheritance
- 8. Staying Engaged with Pogil Packet Answers Inheritance
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pogil Packet Answers Inheritance
- 9. Balancing eBooks and Physical Books Pogil Packet Answers Inheritance
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pogil Packet Answers Inheritance
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pogil Packet Answers Inheritance
 - Setting Reading Goals Pogil Packet Answers Inheritance
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pogil Packet Answers Inheritance

- Fact-Checking eBook Content of Pogil Packet Answers Inheritance
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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