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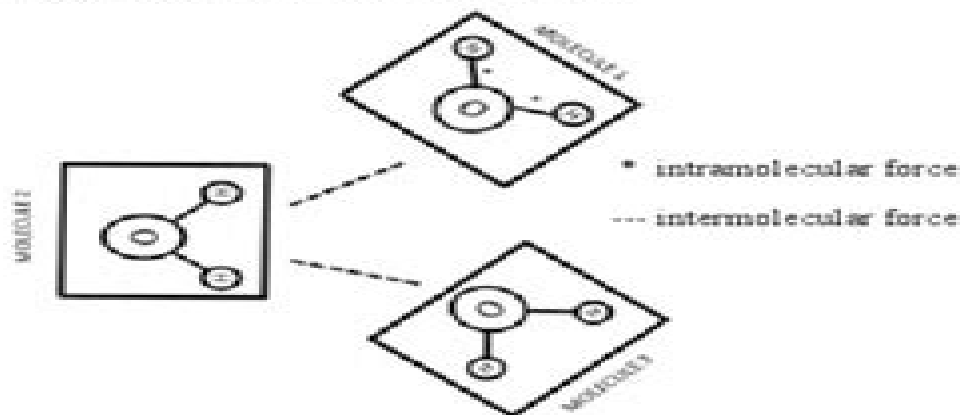
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POGIL: Intermolecular Forces

Model 1: What is an intermolecular force?

As you have learned, matter is made up of discrete particles called **atoms**, which chemically combine to form **molecules**. Molecules do not exist as independent units; in fact, groups of molecules "stick together" in order to form liquids and solids. The forces that hold groups of molecules together are **intermolecular forces**. Without intermolecular forces, the world as we know it would not be the same.

Figure 1: Intramolecular and Intermolecular Forces



Critical Thinking Questions:

1. What specific molecule is represented inside each box in Figure 1?
2. In relation to the box for molecule 1, where do the intramolecular forces exist in Figure 1 – inside the boxes or outside the boxes?
3. Based on the intramolecular forces for molecule 1, draw similar asterisks (*) for the intramolecular forces on the diagram for molecules 2 and 3.
4. In relation to the molecule, where do intramolecular forces tend to occur – within the molecule or outside of the molecule?
5. Two intermolecular forces exist in Figure 1. Where are they positioned relative to the molecules – within the molecules or between the molecules?
6. State the difference between intermolecular and intramolecular forces in terms of where they occur on the molecular level.



Pogil Intermolecular Forces Key

Richard Samuel Moog



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Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes *Intermolecular Forces* Geoffrey C. Maitland, 1981

Intermolecular Forces Pierre L. Huyskens, Werner A.P. Luck, Therese Zeegers-Huyskens, 2012-12-06 The study of intermolecular forces began over one hundred years ago in 1873 with the famous thesis of van der Waals In recent decades knowledge of this field has expanded due to intensive research into both its theoretical and the experimental aspects This is particularly true for the type of very strong cohesive force stressed in 1920 by Latimer and Rodebush the hydrogen bond a phenomenon already outlined in 1912 by Moore and Winemill Hydrogen bonds exert a profound influence on most of the physical and chemical properties of the materials in which they are formed Not only do they govern viscosity and electrical conductivity they also intervene in the chemical reaction path which determines the kinetics of chemical processes The properties of chemical substances depend to a large extent on intermolecular forces In spite of this fundamental fact too little attention is given to these properties both in research and in university teaching For instance in the field of pharmaceutical research about 13000 compounds need to be studied in order to find a single new product that can be successfully marketed The recognition of the need to optimize industrial research efficiency has led to a growing interest in promoting the study of inter molecular forces Rising salary costs in industry have encouraged an interest in theoretical ideas which will lead to tailor made materials **Theory of Intermolecular Forces** H. Margenau, N. R. Kestner, 2013-10-22 Theory of

Intermolecular Forces deals with the exposition of the principles and techniques of the theory of intermolecular forces The text focuses on the basic theory and surveys other aspects with particular attention to relevant experiments The initial chapters introduce the reader to the history of intermolecular forces Succeeding chapters present topics on short intermediate and long range atomic interactions properties of Coulomb interactions shape dependent forces between molecules and physical adsorption The book will be of good use to experts and students of quantum mechanics and advanced physical chemistry **Intermolecular Forces**, Theory of Intermolecular Forces Henry Margenau, N. R. Kestner, 1969

Intermolecular Forces A. Pullman, 2014-01-15 Intermolecular Forces A. Pullman, 2013-06-29 Proceedings of the 14th Jerusalem Symposium on Quantum Chemistry and Biochemistry Jerusalem Israel April 13 16 1981 **Intermolecular Forces** Taro Kihara, 1976 **Convergence of Intermolecular Force Series** Glenn M. Roe, 1952 INTERMOLECULAR

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Intermolecular Forces Tarō Kihara,1978 **The Theory of Intermolecular Forces** Anthony J. Stone,1996 Describes advances in the theory of intermolecular forces and sets out the mathematical techniques that are needed to handle the more elaborate models that are being used increasingly by both theorists and experimentalists Includes a detailed account of the use of higher rank multipole moments to describe electrostatic interactions including treatment of both Cartesian and spherical tensor methods Modern ab initio perturbation theories of intermolecular interactions are also described Annotation copyright by Book News Inc Portland OR **Intermolecular and Surface Forces** Jacob N. Israelachvili,1991

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