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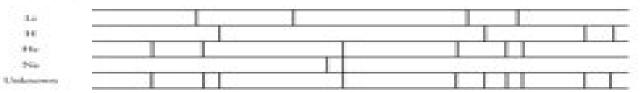


Extension Questions

- 24. The hydrogen spectral lines in Model 2 are only the wavelengths of light that are in the visible range and therefore "seen" by the naked eye. However, many other wavelengths can be detected with special equipment.
 - a. Propose a hydrogen electron transition that involves light with a wavelength in the ultraviolet (UV) range (10—600 nm).

 Propose a hydrogen electron transition that involves light with a wavelength in the infrared (IR) range (1000–106 nm).

 Below are diagrams for the beight line spectra of four elements and the spectrum of a mixture of unknown gases.



a. Which element(s) are not present in the Unknown?

Elements that are metals aren't in the untirum

is. Which element(s) are in the Unknown?

The noble gases are in the waterum.

26. Model 2 shows the emission spectra for hydrogen and boron. Scientists can also record the absorption spectra for elements. Propose how this might be done, and what the absorption spectra of hydrogen and boron would look like.

The light Spectrum of Arce elements could be displayed by projecting light through a prism. The Spectrum of Ocean would be more complex and have many different colors dum the spectrum of hydrogen.

Pogil Electron Energy And Light Extension Questions Answers

Rachel Sandford

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