

Empirical and Molecular Formulas Practice

Read the question carefully for the data provided. Answers must be shown for credit.

Find empirical formula:

1. A compound is found to contain 40.0 g of C and 6.67 g of H. Find the empirical formula.

$$40.0 \text{ g C} \times \frac{1 \text{ mol C}}{12.01 \text{ g C}} = 3.33 \text{ mol C}$$

$$6.67 \text{ g H} \times \frac{1 \text{ mol H}}{1.01 \text{ g H}} = 6.60 \text{ mol H}$$

C₃H₆

2. A compound is found to contain 50.0 g of Fe, 22.0 g of S, and 28.0 g of O. Find the empirical formula.

$$50.0 \text{ g Fe} \times \frac{1 \text{ mol Fe}}{55.85 \text{ g Fe}} = 0.895 \text{ mol Fe}$$

$$22.0 \text{ g S} \times \frac{1 \text{ mol S}}{32.07 \text{ g S}} = 0.686 \text{ mol S}$$



$$28.0 \text{ g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 1.75 \text{ mol O} \left(\frac{0.686 \text{ mol S}}{1.75 \text{ mol O}} = 0.392 \text{ S} = 1 \text{ S} \right)$$

3. A 10.0 g sample of a compound, a gas, contains 2.50 g of C, 3.75 g of H, and 3.75 g of O. Find the empirical formula.

$$2.50 \text{ g C} \times \frac{1 \text{ mol C}}{12.01 \text{ g C}} = 0.208 \text{ mol C}$$



$$3.75 \text{ g H} \times \frac{1 \text{ mol H}}{1.01 \text{ g H}} = 3.71 \text{ mol H} \left(\frac{0.208 \text{ mol C}}{3.71 \text{ mol H}} = 0.056 \text{ C} = 1 \text{ C} \right)$$

$$3.75 \text{ g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 0.234 \text{ mol O} \left(\frac{0.208 \text{ mol C}}{0.234 \text{ mol O}} = 0.89 \text{ C} = 1 \text{ C} \right)$$

4. A 25.00 g sample of an unknown compound contains 7.34 g of C, 9.34 g of H, and 8.32 g of O. Find the empirical formula.

$$7.34 \text{ g C} \times \frac{1 \text{ mol C}}{12.01 \text{ g C}} = 0.611 \text{ mol C}$$



$$9.34 \text{ g H} \times \frac{1 \text{ mol H}}{1.01 \text{ g H}} = 9.24 \text{ mol H}$$

$$8.32 \text{ g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 0.520 \text{ mol O} \left(\frac{0.611 \text{ mol C}}{0.520 \text{ mol O}} = 1.17 \text{ C} = 1 \text{ C} \right)$$

Pogil Chemistry Answer Key Empirical

Lingsheng Yao



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