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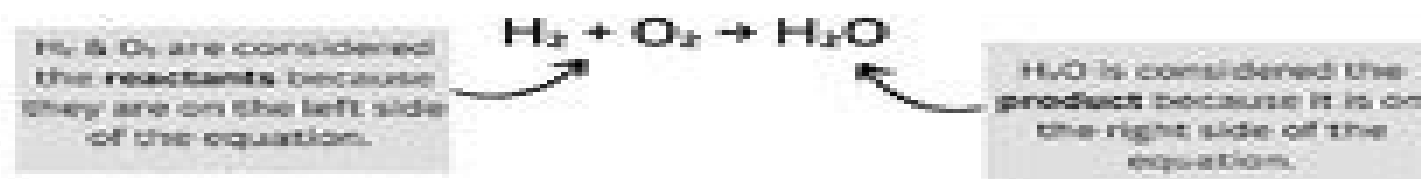
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Balancing Chemical Reactions POGIL

Steps to balancing a chemical reaction:

1. **Writing the Chemical Equation** - Balancing a chemical reaction always starts with writing it out first! A chemical equation shows the **reactants** on the **left side** and the **products** on the **right side** of the reaction arrow.

For example, let's consider the reaction between hydrogen gas (H_2) and oxygen gas (O_2) to produce water (H_2O):

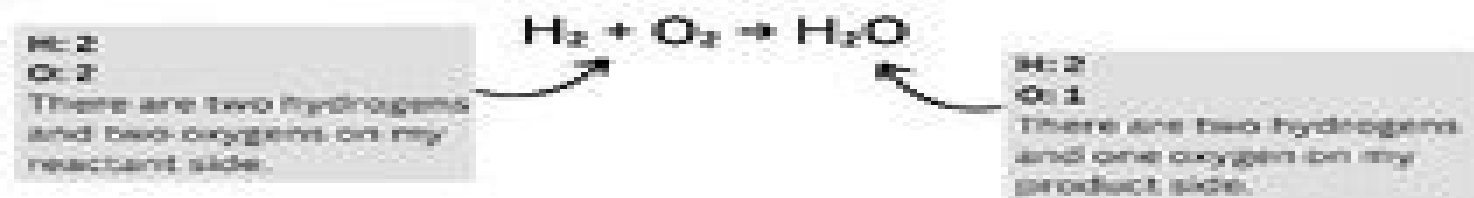


Try identifying the reactants and products of the following chemical equations:



2. **Count the Number of Atoms on Each Side** - Make a list of all atoms on both the reactant side and product side (you will have the same atoms even if they look different). Count individually how many of each atom there are.

For example, let's again consider the reaction between hydrogen gas (H_2) and oxygen gas (O_2) to produce water (H_2O):



Pogil Chemical Equations

Chao Zhang



Pogil Chemical Equations:

Pogil Chemical Equations Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Pogil Chemical Equations**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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Pogil Chemical Equations Introduction

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