

Bond Energy

What makes a reaction endothermic or exothermic?

Why?

Chemical reactions can be a lot like playing with Legos®—you must take apart part of your last creation before you can replace it with something new. For many chemical reactions, we have to first break bonds in the reactants before we can put the atoms back into a new arrangement to form the products. Both of these processes involve changes in energy. The net energy change for a reaction is called the **heat of reaction** or the **change in enthalpy** (ΔH). In this activity we will look at one way energy changes can be approximated for chemical reactions.

Model 1 – Breaking and Forming Bonds

	Reaction	Change in Energy (kJ/mol _{rxn})		Reaction	Change in Energy (kJ/mol _{rxn})
A)	$\text{PCl}_3(\text{g}) \rightarrow \text{P}(\text{g}) + 3\text{Cl}(\text{g})$	+ 966.7 <i>broken</i>	E)	$\text{P}(\text{g}) + 3\text{Cl}(\text{g}) \rightarrow \text{PCl}_3(\text{g})$	- 966.7 <i>formed</i>
B)	$\text{PCl}_5(\text{g}) \rightarrow \text{P}(\text{g}) + 5\text{Cl}(\text{g})$	+ 1,297.9 <i>broken</i>	F)	$\text{P}(\text{g}) + 5\text{Cl}(\text{g}) \rightarrow \text{PCl}_5(\text{g})$	- 1,297.9 <i>formed</i>
C)	$\text{PF}_3(\text{g}) \rightarrow \text{P}(\text{g}) + 3\text{F}(\text{g})$	+ 1,470.4 <i>broken</i>	G)	$\text{P}(\text{g}) + 3\text{F}(\text{g}) \rightarrow \text{PF}_3(\text{g})$	- 1,470.4 <i>formed</i>
D)	$\text{PF}_5(\text{g}) \rightarrow \text{P}(\text{g}) + 5\text{F}(\text{g})$	+ 2,305.4 <i>broken</i>	H)	$\text{P}(\text{g}) + 5\text{F}(\text{g}) \rightarrow \text{PF}_5(\text{g})$	- 2,305.4 <i>formed</i>

- Identify four reactions in Model 1 where bonds are being broken. Write "broken" in the Change in Energy box in the table for each reaction you identify.
- Identify four reactions in Model 1 where bonds are being formed. Write "formed" in the Change in Energy box in the table for each reaction you identify.
- Circle the correct word to complete each sentence below.
 - When bonds are (broken/formed) there is a positive energy change.
 - Breaking bonds is (endothermic/exothermic).
 - When bonds are (broken/formed) there is a negative energy change.
 - Forming bonds is (endothermic/exothermic).
- Find two reactions in Model 1 that are exact opposites of each other, that is, one reaction is the reverse of the other reaction.

- How do the changes in energy for the reverse reactions compare?

The changes in energy for these reactions are the same, but opposite in sign.

- Explain your answer to part a considering what you learned from Questions 1–3 about bond breaking and bond formation.

If a given amount of energy is required to break a bond, then the same amount of energy will be released when the bond is formed.

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S Nieto



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web than countries with lower entrepreneurial activity one third of this difference could be traced directly back to entrepreneurs the most important reason why entrepreneurs help the economy is due to the creation of new jobs it has been

estimated that entrepreneurs have created 34 million new jobs since 1980 also 70 percent of new

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web 1 1 unit 1 introduction and learning objectives 1 2 defining entrepreneurship 1 3 the role entrepreneurs have in today s society 1 4 different types of entrepreneurship

unit 2 entrepreneurial characteristics introduction to entrepreneurship - Feb 03 2022

web learning outcomes reflect on entrepreneurial skills and abilities analyze what leads to entrepreneurial success debate the role of creativity and innovation in entrepreneurship previous different types of entrepreneurship next

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web according to zimmerer and scarborough entrepreneurship is the creation of a new business in the face of risk and uncertainty for the purpose of achieving profit and growth by indentifying significant opportunities and assembling the necessary resources to

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web the pursuit of opportunity beyond the resources you currently control study with quizlet and memorize flashcards containing terms like entrepreneurship for social inclusion seeks growth by tv and internet have provided some valuable education on the topic of entrepreneurship t f for entrepreneurship to thrive it must operate in a well

bus305 introduction to entrepreneurship saylor academy - Jan 14 2023

web nov 4 2020 introduction to entrepreneurship at the end of this article you will be able to define entrepreneurship within the context of standard activities and organizational support entrepreneurs are innovators willing to take risks and generate new ideas to create unique and potentially profitable solutions to modern day problems

[1 1 unit 1 introduction and learning objectives](#) - Jun 19 2023

web in this unit we are going to learn about the nature of entrepreneurship and why it matters entrepreneurship is a vast concept that is often misunderstood to mean something very narrow when in fact it is quite broad and can be interpreted in many ways

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web 1 0 introduction this unit is designed to provide an insight into the definitions nature and concept of entrepreneurship by the end of the unit you are expected to have an overview of what entrepreneurship entails functions and the difference s if any between entrepreneurship and intrapreneurship

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web may 12 2023 i unit 1 what is entrepreneurship 1 defining entrepreneurship 2 the role entrepreneurs have in today s society 3 different types of entrepreneurship ii unit 2 entrepreneurial characteristics 4 entrepreneurial traits skills and

abilities 5 the entrepreneurial mindset 6 creativity and innovation in entrepreneurship iii

defining entrepreneurship introduction to entrepreneurship - Nov 12 2022

web the entrepreneur the entrepreneur theme is the idea that entrepreneurship involves individuals with unique personality characteristics and abilities e g risk taking locus of control autonomy perseverance commitment vision creativity

module 1 introduction to entrepreneur and entrepreneurship - Aug 09 2022

web module 1 introduction to entrepreneur and entrepreneurship topics to be covered meaning of entrepreneur evolution of the concept functions of an entrepreneur types of entrepreneur intrapreneur an emerging class concept of entrepreneurship entrepreneurial culture stages in entrepreneurial process