

Write a balance equation and label the states of the reactants and products for each in each reaction. After writing the balanced equation, write the total ionic equation (TIE), list the driving force (DF) and spectator ions (SI), and write the net ionic equation(s) (NET) below the balanced equation. If there is no driving force for a reaction, write no reaction in place of the TIE, DF, SI, and NET.

1. Hydrogen sulfide is bubbled through a solution of strontium hydroxide.



2. Aluminum metal is added to a solution of copper (II) chloride.



3. A manganese (II) nitrate solution is mixed with a sodium hydroxide solution.



Pogil Activity Solubility Rules And Net Ionic Equations Answers

K Morrison



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