

Mole-Mole Stoichiometry Worksheet

Complete each of the following mole to mole stoichiometric conversions.



How many moles of KCl are produced from 2.00 moles of K and excess Cl_2 ?



How many moles of iron are needed to react with 16.0 moles of sulfur?



How many moles of Cu are needed to react with 3.00 moles of AgNO_3 ?



How many moles of K are needed to react with 0.5 moles of Cl_2 ?



How many moles of Fe are needed to react with 1.00 moles of O_2 ?



How many moles of zinc are needed to make 6 moles of zinc oxide?



5.00 moles of hydrogen react with excess oxygen. What is the maximum number of moles of water that can be formed?



4 moles of oxygen react with excess nitrogen. What is the maximum number of moles of dinitrogen tetroxide (N_2O_4) that can be produced?



How many grams of KCl are produced from 78.20 g of K and excess Cl_2 ?

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