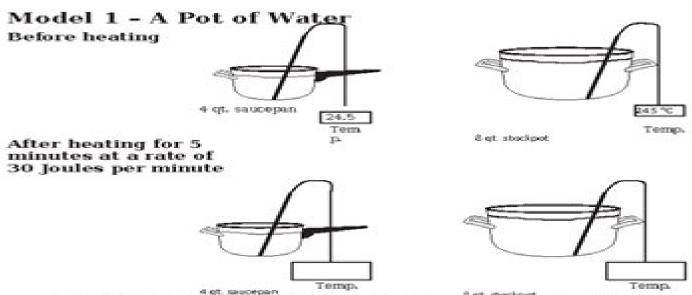
Calorimetry

What is the relationship between heat energy and temperature?

Why?

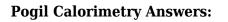
When a substance is heated, the temperature of that substance increases. Will the same amount of energy cause different substances to have identical temperature increases? Will the same amount of energy be needed to cause identical temperature increases in different amounts of the same substance? In this activity you will explore how mass, temperature, heat energy, and the type of substance are related.



- 1. In Model 1, which container holds more grams of water?
- 2. Consider the process described in Model 1:
 - a. How many joules of energy were added to the saucepan?
 - b. How many joules of energy were added to the stockpot?
 - c. In which container did the liquid gain more energy or did both gain the same amount? Explain your reasoning.

Pogil Calorimetry Answers

G Orfield



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