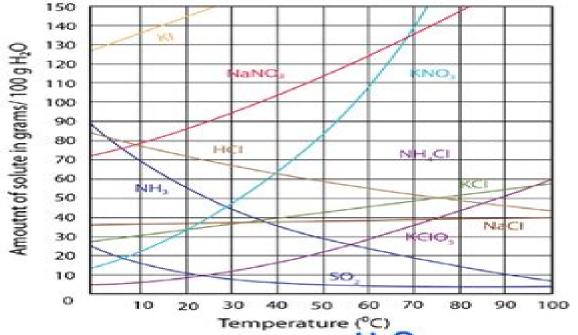
Interpreting Solubility Curve 1

- Define solubility. The maximum amount of solute that can dissolve
 Using the solubility curve below, answer the given questions.



I. What is the solvent in this solubility curve? H_2O

10 II. Saturation points are given per how much of the solvent?

III. What is the saturation point of NaNO3 at 10°C, 40°C, & 80°C?

V. Which salt is the least soluble in water? NaCl

VI. How many grams of potassium chloride can be dissolved in 200 g of water at 80 °C?

VII. At 40 °C, how much potassium nitrate will be dissolved in 300 g of water?

VIII. Which salt shows the least change in solubility from 0° to 100° C?

IX. Which substance shows a decrease in solubility from 0° to 100° C?

X. At 30 °C, 90 g of sodium nitrate is dissolved in 100 g water. Is this solution saturated, unsaturated, or supersaturated?

Pogil Answer Key To Interpreting Solubility Curves

S Ashworth

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