

Name: \_\_\_\_\_

# CLASSIFYING CHEMICAL REACTIONS

Classify the following chemical reactions as one of the following: synthesis, decomposition, single-replacement, double-replacement, or combustion.

|                                                                                |                                                                                    |
|--------------------------------------------------------------------------------|------------------------------------------------------------------------------------|
| $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$                     | $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$                               |
| $2\text{K} + 2\text{H}_2\text{O} \rightarrow \text{H}_2 + 2\text{KOH}$         | $\text{CaCO}_3 + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2\text{CO}_3$    |
| $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$                         | $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$                   |
| $2\text{SO}_2 + \text{O}_2 \rightarrow 2\text{SO}_3$                           | $\text{Pb}(\text{NO}_3)_2 + 2\text{HCl} \rightarrow \text{PbCl}_2 + 2\text{HNO}_3$ |
| $\text{Cu} + 2\text{AgNO}_3 \rightarrow 2\text{Ag} + \text{Cu}(\text{NO}_3)_2$ | $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$                                |
| $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$                     | $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$          |
| $\text{Pb}(\text{NO}_3)_2 + 2\text{K} \rightarrow \text{Pb} + 2\text{KNO}_3$   | $\text{H}_2\text{CO}_3 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$               |
| $2\text{Fe} + \text{O}_2 \rightarrow 2\text{FeO}$                              | $\text{Zn} + 2\text{HCl} \rightarrow \text{H}_2 + \text{ZnCl}_2$                   |
| $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$        | $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$           |

# Physical Science If8767 Classifying Chemical Reactions

**Joacim Rocklöv**



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## **Table of Contents Physical Science If8767 Classifying Chemical Reactions**

1. Understanding the eBook Physical Science If8767 Classifying Chemical Reactions
  - The Rise of Digital Reading Physical Science If8767 Classifying Chemical Reactions
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Classifying Chemical Reactions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science If8767 Classifying Chemical Reactions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Classifying Chemical Reactions

- Personalized Recommendations
  - Physical Science If8767 Classifying Chemical Reactions User Reviews and Ratings
  - Physical Science If8767 Classifying Chemical Reactions and Bestseller Lists
5. Accessing Physical Science If8767 Classifying Chemical Reactions Free and Paid eBooks
    - Physical Science If8767 Classifying Chemical Reactions Public Domain eBooks
    - Physical Science If8767 Classifying Chemical Reactions eBook Subscription Services
    - Physical Science If8767 Classifying Chemical Reactions Budget-Friendly Options
  6. Navigating Physical Science If8767 Classifying Chemical Reactions eBook Formats
    - ePub, PDF, MOBI, and More
    - Physical Science If8767 Classifying Chemical Reactions Compatibility with Devices
    - Physical Science If8767 Classifying Chemical Reactions Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Physical Science If8767 Classifying Chemical Reactions
    - Highlighting and Note-Taking Physical Science If8767 Classifying Chemical Reactions
    - Interactive Elements Physical Science If8767 Classifying Chemical Reactions
  8. Staying Engaged with Physical Science If8767 Classifying Chemical Reactions
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Physical Science If8767 Classifying Chemical Reactions
  9. Balancing eBooks and Physical Books Physical Science If8767 Classifying Chemical Reactions
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Physical Science If8767 Classifying Chemical Reactions
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Physical Science If8767 Classifying Chemical Reactions
    - Setting Reading Goals Physical Science If8767 Classifying Chemical Reactions
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Physical Science If8767 Classifying Chemical Reactions

- Fact-Checking eBook Content of Physical Science If8767 Classifying Chemical Reactions
- Distinguishing Credible Sources

### 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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