

TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

1	Which of the following is a uni-univalent electrolyte? a) KCl b) CuCl_2 c) AlCl_3 d) MgCl_2
2	The ionic strength of a solution of molality m_i and valency of the ion Z_i is ____ a) $\mu = \frac{1}{2} m_i Z_i^2$ b) $\mu = \sum m_i Z_i^2$ c) $\mu = \frac{1}{2} \sum Z_i^2$ d) $\mu = \frac{1}{2} \sum m_i Z_i^2$
3	Which of the following is an example of electrolyte concentration cell with transference? a) $\text{Cu} \text{CuSO}_4 \text{AgNO}_3 \text{Ag}$ b) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)} \text{AgCl}_{(s)} \text{Ag}$ c) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)}(a_1) \text{AgNO}_3(a_2) \text{Ag}$ d) $\text{Ag} - \text{AgCl} \text{NaCl}(a_1) \text{NaCl}(a_2) \text{AgCl} \cdot \text{Ag}$
4	A concentration cell which has salt bridge in its representation is ____ a) electrolyte concentration cell with transference b) Electrode concentration cell c) Electrolyte concentration cell without transference d) Gas concentration cell
5	The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by..... a) $E_j = E_{\text{oxidation}} + E_{\text{reduction}}$ b) $E_{\text{cell}} = E_{\text{oxidation}} + E_{\text{reduction}} + E_j$ c) $E_{\text{cell}} = E_{\text{oxidation}} - E_{\text{reduction}} - E_j$ d) $E_j = E_{\text{oxidation}} - E_{\text{reduction}}$
6	The activity 'a' is related to the activity coefficient ' γ ' by the equation____ a) $a = \frac{m}{\gamma}$ b) $\gamma = a - m$ c) $a = m + \gamma$ d) $a = m\gamma$

Physical Chemistry Multiple Choice Questions

M.M.L. Oblitas



Physical Chemistry Multiple Choice Questions:

This book delves into Physical Chemistry Multiple Choice Questions. Physical Chemistry Multiple Choice Questions is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Physical Chemistry Multiple Choice Questions, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Physical Chemistry Multiple Choice Questions
 - Chapter 2: Essential Elements of Physical Chemistry Multiple Choice Questions
 - Chapter 3: Physical Chemistry Multiple Choice Questions in Everyday Life
 - Chapter 4: Physical Chemistry Multiple Choice Questions in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Physical Chemistry Multiple Choice Questions. The first chapter will explore what Physical Chemistry Multiple Choice Questions is, why Physical Chemistry Multiple Choice Questions is vital, and how to effectively learn about Physical Chemistry Multiple Choice Questions.
 3. In chapter 2, the author will delve into the foundational concepts of Physical Chemistry Multiple Choice Questions. The second chapter will elucidate the essential principles that need to be understood to grasp Physical Chemistry Multiple Choice Questions in its entirety.
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 6. In chapter 5, this book will draw a conclusion about Physical Chemistry Multiple Choice Questions. This chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Physical Chemistry Multiple Choice Questions.

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