

## Periodic Trends Worksheet

Use the periodic table and your knowledge of periodic trends to answer the following questions.

1. Which atom in each pair has the larger atomic radius?  
a) Li or **K**      b) **Ca** or Na      c) **Ga** or B      d) O or **C**      e) Cl or **Br**  
f) Be or **Ba**      g) **Si** or S      h) Fe or **Au**

2. What is the periodic trend for atomic size from top to bottom in a group? from left to right in a period?

Top to bottom: increases due to adding energy levels    Left to Right: decreases

3. Why do atoms get smaller as you move left to right in a period?  
due to increased nuclear charge and no added shielding

4. Which element in each pair has a larger ionization energy?  
a) Na or **O**      b) **Be** or Ba      c) **Ar** or F      d) **Cu** or Ra      e) I or **Ne**  
f) K or **V**      g) **Ca** or Fr      h) W or **Se**

5. Explain the relationship between the relative size of an ion to its neutral atom and the charge on the ions.  
Positive ions are smaller than the neutral atom it comes from, negative ions are larger than the neutral atom it comes from

6. Which particle has the larger radius in each atom/ion pair?  
a) Na,  $\text{Na}^+$       b) S,  $\text{S}^{2-}$       c) I,  $\text{I}^-$       d) **Al**,  $\text{Al}^{3+}$

7. What is ionization energy? What is first ionization energy?

The amount of energy needed to remove an electron from an atom. First ionization energy is the energy required to remove a single valence electron from an atom.

8. What is the periodic trend for first ionization energy?

It increases going from left to right due to increased nuclear charge, and decreases going down a group due to increased distance from the nucleus and increased shielding.

9. Arrange the following groups of elements in order of increasing ionization energy.  
a) Be, Mg, Sr      b) Bi, Cs, Ba      c) Na, Al, S  
high, middle, low      high, low, middle      low, middle, high

10. Which element in each pair has a higher electronegativity value?  
a) Cl, **F**      b) C, **N**      c) **Mg**, Ne      d) As, Ca

11. What is the periodic trend for electronegativity? Increases going left to right, decreases going down.

# Periodic Trends Pre Lab Answers

**Xiang Xie**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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**Sustainable Green Chemistry** Mark Anthony Benvenuto, 2017-04-10 Sustainable Green Chemistry the 1st volume of Green Chemical Processing covers several key aspects of modern green processing The scope of this volume goes beyond bio and organic chemistry highlighting the ecological and economic benefits of enhanced sustainability in such diverse fields as petrochemistry metal production and wastewater treatment The authors discuss recent progresses and challenges in the implementation of green chemical processes as well as their transfer from academia to industry and teaching at all levels Selected successes in the greening of established processes and reactions are presented including the use of switchable polarity solvents actinide recovery using ionic liquids and the removal of the ubiquitous bisphenol A molecule from effluent streams by phytodegradation

**Integrated Approach to Coordination Chemistry** Rosemary A. Marusak, Kate Doan, Scott D. Cummings, 2007-03-07 Coordination chemistry is the study of compounds formed between metal ions and other neutral or negatively charged molecules This book offers a series of investigative inorganic laboratories approached through systematic coordination chemistry It not only highlights the key fundamental components of the coordination chemistry field it also exemplifies the historical development of concepts in the field In order to graduate as a chemistry major that fills the requirements of the American Chemical Society a student needs to take a laboratory course in inorganic chemistry Most professors who teach and inorganic chemistry laboratory prefer to emphasize coordination chemistry rather than attempting to cover all aspects of inorganic chemistry because it keeps the students focused on a cohesive part of inorganic chemistry which has applications in medicine the environment molecular biology organic synthesis and inorganic materials

**Guided Inquiry Experiments for General Chemistry** Nancy K. Kerner, Ram S. Lamba, 2007-10-19 The use of the laboratory is a valuable tool in developing a deeper understanding of key chemical concepts from the experimental process This lab manual encourages scientific thinking enabling readers to conduct investigations in chemistry It shows how to think about the processes they are investigating rather than simply performing a laboratory experiment to the specifications set by the manual Each experiment begins with a problem scenario and ends with questions requiring feedback on the problem *Laboratory Manual for Principles of General Chemistry Jo Allan*

Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures **The World of Chemistry** Jones,Jones Staff,1996

Science Spectrum Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2003-03 **Exploring Physical Science in the Laboratory** John T. Salinas ,2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts *Diagnostic Pathology: Breast - Ebook* Susan C. Lester,David G. Hicks,2025-05-06 This expert volume in the Diagnostic Pathology series is not only an up to date comprehensive diagnostic support tool for surgical pathologists but also a valuable resource for all health care workers who provide care to patients with breast disease including radiologists as well as surgical medical and radiation oncologists An excellent point of care reference for practitioners at all levels of experience and training the fourth edition of Diagnostic Pathology Breast Fourth Edition provides details on normal breast histology information to assist in processing breast specimens and multichapter sections on diagnostic patterns benign lesions carcinomas predictive and prognostic factors stromal lesions inflammatory lesions lymphomas and hereditary breast disease Richly illustrated and easy to use this volume is ideal as a one stop resource for day to day reference or as a reliable training resource Incorporates the most up to date scientific and technical knowledge providing a comprehensive overview of all key issues relevant to today s practice Emphasizes the correlation of pathologic lesions with findings on breast imaging and provides guidance on the interpretation of core needle biopsies Contains new overview chapters that lay the groundwork for initial evaluation and assistance with differential diagnosis as well as new chapters on HER2 low breast carcinoma immunohistochemical studies for diagnosis proliferation and more Provides important updates throughout covering the evaluation of ESR1 PIK3CA AKT and BRCA1 mutations to guide therapy tumor infiltrating lymphocytes and PD L1 for treatment planning HER2 low assessment to determine eligibility for antibody drug conjugates and much more Features new animations and new videos to assist in understanding three dimensional anatomy radiologic pathologic correlation the processing of breast specimens and the gross identification of breast lesions Contains more than 3 700 extensively annotated images including gross pathology photographs histopathology photomicrographs a wide range of immunohistochemical stains fluorescent in situ hybridization breast imaging studies and full color illustrations Employs consistently templated

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*Energy Research Abstracts*, 1992 **Applied Mechanics Reviews**, 1982 **Selected Water Resources Abstracts**, 1989 **Quality Control and Regulatory Aspects for Biologicals** Gauri Misra, 2024-04-22 This book serves as a comprehensive guide on quality control and regulatory aspects for biological products It covers a wide range of topics including regulatory requirements quality control strategies analytical methods and risk management It delves into the advantages and limitations of in vivo tests and discusses alternative methods that can be employed The book explores the use of animal based testing methods in quality control and examines viable alternatives Key Features Reviews various scientific and regulatory aspects involved in the quality control of biologicals Provides an overview of the roles of various national and international regulatory bodies and accreditation agencies Presents advanced analytical methods innovative technologies and the integration of molecular diagnostics in quality control processes Explores the use of animal based testing methods in quality control as well as their alternatives Discusses guidelines and methodologies involved in the development of biological products Overall this book is an important reference source for various professionals in the pharmaceutical industry including researchers scientists quality control personnel and regulatory affairs professionals

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