

# PERIODIC TABLE

## Atomic Properties of the Elements

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Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data					
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1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of <sup>133</sup> Cs																									
speed of light in vacuum c = 299 792 458 m s <sup>-1</sup> (exact)																									
Planck constant h = 6.626 070 15 × 10 <sup>-34</sup> J s (exact)																									
elementary charge e = 1.602 176 634 × 10 <sup>-19</sup> C																									
electron mass m <sub>e</sub> = 9.109 383 71 × 10 <sup>-31</sup> kg																									
proton mass m <sub>p</sub> = 1.672 621 9 × 10 <sup>-27</sup> kg																									
fine-structure constant α = 1/137.036																									
Rydberg constant R <sub>∞</sub> = 10 973 731.568 162 m <sup>-1</sup>																									
Bohr magneton μ <sub>B</sub> = 9.274 009 474 × 10 <sup>-24</sup> J T <sup>-1</sup>																									
Schrödinger constant μ <sub>N</sub> = 5.050 646 218 × 10 <sup>-27</sup> J T <sup>-1</sup>																									

Atomic Number: 58  
Ground-state Level:  $4f^7 5s^1$

Symbol: **Ce**

Name: Cerium

Atomic Weight: 140.116  
([Xe]4f<sup>7</sup>5s<sup>1</sup>)

Ground-state Configuration: [Xe]4f<sup>7</sup>5s<sup>1</sup>

Ionization Energy (eV): 5.5387

<sup>1</sup>Based upon <sup>12</sup>C. ( ) indicates the mass number of the longest-lived isotope.

For a description of the data, visit [physics.nist.gov/data](http://physics.nist.gov/data)

# Periodic Table Chemistry If8766

**Xiaolong Qi**



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