Using the Periodic Table

C - Introduction to the Periodic Table

d) a. Using the blank Periodic Table provided 6 your understanding of atomic number, describe how

this identicate an list of on the Resolution table.

- b. Using the Periodic Table to the right, draw a circle around a "once", and fill in one "period".
- c. Based on the evidence we have gathered from scientific research, what do we know about elements that are found in the same.

group on the Periodic Table?

1		911										93			
1		8									н				
1															П
1			-	-	H	-	Н		-	H	Н		-	н	Н
4								23					*		

Description of the Control of the Co	of the second second	Annual Control of the	The second second second
	The second second		

70. Give the symbol and atomic

number for each of the following elements and indicate whether each element is a metal, or a normatal, and whether they are members of a named proportionally on the Penadic Table.

Harrie	Symbol	Abomic	Hetal or Honometal	Group/Family Mame
Argon				
Chicomine				
Stanium.				
Cerenum				

- 60 Based on your knowledge of the Periodic Table and the natural states that elements occur in.
 - a. An element referred to as a noble gas -
 - b. An element that naturally occurs as a distornic molecule -
 - An element that occurs as a solid at room temperature -
 - d. An element that occurs as a liquid at room temperature -...
 - An element that occurs as a gas at room temperature -

tions & Their Compounds

90 Wifey does an aborn have a zero net charge?

10) tons are formed by an atom gaming or losing

11) Metals naturally form what type of ion? Bionmetals? Provide definitions for each type of ion.

320 Describe what happens to:

- a. the size of an atom when it loses an electron
- b. the size of an atom when it gains an electron

<u>Periodic Table Chemistry Guided Practice Problems</u> <u>Answers</u>

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