

## Test Periodic Table and Trends SCH3U

Name: \_\_\_\_\_

Multiple Choice: Select the most correct answer (20K)

1. The mass number of an atom is determined by:

- a. the number of protons only
- b. adding the neutrons and protons
- c. adding the neutrons and electrons
- d. adding the protons and electrons



2. The letter "X" in the diagram above marks the:

- a. Region of greatest density
- b. Location of the protons
- c. Location of the neutrons
- d. Location of the electrons
- e. Nucleus

3. Most of the mass of the atom can be found in the:

- a. charges    b. electron cloud    c. electrons    d. nucleus

4. Of the basic atomic particles, the one that would be attracted to a negatively charged metallic plate is the:

- a. proton    b. neutron    c. electron    d. none of the above

5. Isotopes (such as hydrogen-1, hydrogen-2, and hydrogen-3) are atoms of the same element that differ in:

- a. the number of protons in the nucleus
- b. the number of electrons
- c. the atomic number
- d. the number of neutrons in the nucleus
- e. the mass number

6. Which of the following accurately summarizes the isotope Argon-40?

- a. 40 protons, 40 electrons, 18 neutrons
- b. 18 protons, 18 electrons, 22 neutrons
- c. 39.95 protons, 39.95 electrons, 21.05 neutrons
- d. 22 protons, 22 electrons, 18 neutrons

# Periodic Table Test Answers

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Silver**



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## **Table of Contents Periodic Table Test Answers**

1. Understanding the eBook Periodic Table Test Answers
  - The Rise of Digital Reading Periodic Table Test Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Test Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Test Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Test Answers
  - Personalized Recommendations
  - Periodic Table Test Answers User Reviews and Ratings
  - Periodic Table Test Answers and Bestseller Lists
5. Accessing Periodic Table Test Answers Free and Paid eBooks
  - Periodic Table Test Answers Public Domain eBooks
  - Periodic Table Test Answers eBook Subscription Services
  - Periodic Table Test Answers Budget-Friendly Options
6. Navigating Periodic Table Test Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table Test Answers Compatibility with Devices
  - Periodic Table Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table Test Answers
  - Highlighting and Note-Taking Periodic Table Test Answers
  - Interactive Elements Periodic Table Test Answers
8. Staying Engaged with Periodic Table Test Answers

- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table Test Answers
9. Balancing eBooks and Physical Books Periodic Table Test Answers
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Test Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Test Answers
- Setting Reading Goals Periodic Table Test Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Test Answers
- Fact-Checking eBook Content of Periodic Table Test Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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