

Percent Composition Practice 1.8 -- Answer Key

**Objective:** - Calculate the percent composition of a compound.

**Directions:** Determine the percent composition of each of the compounds below. (Show all work!)

1.  $\text{HCl}$

$$\text{H: } 1.00\text{g} \times 1 = 1.01\text{g} \quad \left( \frac{1.01\text{g}}{35.46\text{g}} \right) \times 100\% = \boxed{2.73\%}$$

$$\text{Cl: } 35.45\text{g} \times 1 = 35.45\text{g} \quad \left( \frac{35.45\text{g}}{35.46\text{g}} \right) \times 100\% = \boxed{97.27\%}$$

$36.46\text{g}$

2.  $\text{KMnO}_4$

$$\text{K: } 39.10\text{g} \times 1 = 39.10\text{g} \quad \left( \frac{39.10\text{g}}{158.04\text{g}} \right) \times 100\% = \boxed{24.74\%}$$

$$\text{Mn: } 54.94\text{g} \times 1 = 54.94\text{g} \quad \left( \frac{54.94\text{g}}{158.04\text{g}} \right) \times 100\% = \boxed{34.76\%}$$

$$\text{O: } 16.00\text{g} \times 4 = 64.00\text{g} \quad \left( \frac{64.00\text{g}}{158.04\text{g}} \right) \times 100\% = \boxed{40.50\%}$$

$158.04\text{g}$

3.  $\text{Mg(NO}_3)_2$

$$\text{Mg: } 24.31\text{g} \times 1 = 24.31\text{g} \quad \left( \frac{24.31\text{g}}{148.33\text{g}} \right) \times 100\% = \boxed{16.39\%}$$

$$\text{N: } 14.01\text{g} \times 2 = 28.02\text{g} \quad \left( \frac{28.02\text{g}}{148.33\text{g}} \right) \times 100\% = \boxed{18.89\%}$$

$$\text{O: } 16.00\text{g} \times 6 = 96.00\text{g} \quad \left( \frac{96.00\text{g}}{148.33\text{g}} \right) \times 100\% = \boxed{64.72\%}$$

$148.33\text{g}$

4.  $(\text{NH}_4)_3\text{PO}_4$

$$\text{N: } 14.01\text{g} \times 3 = 42.03\text{g} \quad \left( \frac{42.03\text{g}}{149.12\text{g}} \right) \times 100\% = \boxed{28.19\%}$$

$$\text{H: } 1.01\text{g} \times 12 = 12.12\text{g} \quad \left( \frac{12.12\text{g}}{149.12\text{g}} \right) \times 100\% = \boxed{8.13\%}$$

$$\text{P: } 30.97\text{g} \times 1 = 30.97\text{g} \quad \left( \frac{30.97\text{g}}{149.12\text{g}} \right) \times 100\% = \boxed{20.77\%}$$

$$\text{O: } 16.00\text{g} \times 4 = 64.00\text{g} \quad \left( \frac{64.00\text{g}}{149.12\text{g}} \right) \times 100\% = \boxed{42.92\%}$$

$149.12\text{g}$

# Percent Composition Answer Sheet

**Kaplan**



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## **Table of Contents Percent Composition Answer Sheet**

1. Understanding the eBook Percent Composition Answer Sheet
  - The Rise of Digital Reading Percent Composition Answer Sheet
  - Advantages of eBooks Over Traditional Books
2. Identifying Percent Composition Answer Sheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Percent Composition Answer Sheet
  - User-Friendly Interface
4. Exploring eBook Recommendations from Percent Composition Answer Sheet
  - Personalized Recommendations
  - Percent Composition Answer Sheet User Reviews and Ratings
  - Percent Composition Answer Sheet and Bestseller Lists
5. Accessing Percent Composition Answer Sheet Free and Paid eBooks

- Percent Composition Answer Sheet Public Domain eBooks
- Percent Composition Answer Sheet eBook Subscription Services
- Percent Composition Answer Sheet Budget-Friendly Options
- 6. Navigating Percent Composition Answer Sheet eBook Formats
  - ePub, PDF, MOBI, and More
  - Percent Composition Answer Sheet Compatibility with Devices
  - Percent Composition Answer Sheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Percent Composition Answer Sheet
  - Highlighting and Note-Taking Percent Composition Answer Sheet
  - Interactive Elements Percent Composition Answer Sheet
- 8. Staying Engaged with Percent Composition Answer Sheet
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Percent Composition Answer Sheet
- 9. Balancing eBooks and Physical Books Percent Composition Answer Sheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Percent Composition Answer Sheet
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Percent Composition Answer Sheet
  - Setting Reading Goals Percent Composition Answer Sheet
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Percent Composition Answer Sheet
  - Fact-Checking eBook Content of Percent Composition Answer Sheet
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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