



POCCL-The Mole Concept (How can atoms be counted?)

Notes:

Consider the following representation of a chemical reaction: $2H_2 + O_2 \rightarrow 2H_2O$

This can be interpreted either as two molecules of hydrogen and one molecule of oxygen combining to make two water molecules, OR two atoms of hydrogen and one oxygen atom or two molecules of diatomic hydrogen (these will be discussed later) reacting with one molecule of diatomic oxygen to make two water molecules. The molecules need to be quickly counted! How do we count molecules? The answer is the mole and the mole. Thinking is often easier than counting!

Model 1: The Balanced Equation

How can the coefficients in a chemical equation be interpreted?

Consider the reaction:



Requirements:

1. What can the coefficients each be interpreted as relating to the reaction?

coefficients represent each separate molecule
atoms reacting separately
atoms only!

Model 1.1.1

Coefficients tell you the relative number of products and reactants. A balanced chemical equation can be interpreted in balanced relative quantities including:

- Numbers of particles (atoms, molecules, or formula units)

(e.g. $2H_2O$ is 2)

Atoms

Molecules

Formula Units

(All particles are single elements)

(All particles are molecules/compounds)

All particles have been fully decomposed

- Atoms

- Molecules

Pogil The Mole Concept Key

JG Myers



Pogil The Mole Concept Key:

Exploring the Effectiveness of POGIL and Chemorganisers in Foundation Chemistry Christine Elizabeth Mundy, 2015 The extended BSc programme at the University of Pretoria was the context of this study specifically students enrolled in foundation chemistry This study was aimed at improving teaching and learning at this level by implementing pre existing education interventions Process Oriented Guided Inquiry Learning POGIL and the use of the Chemorganiser Themes chosen for the interventions were the mole concept and stoichiometry and redox reactions as these have been identified as common areas of difficulty in Chemistry Johnston 2010 POGIL required students to take on well defined roles and work in groups on specially designed worksheets Farrell Moog 1990 served as the theoretical lens for the study The effectiveness of each intervention was explored using the classroom variables of productivity student preference and opportunity to learn along with the output of student performance Mixed methods including observations focus group interviews student questionnaires and student performance data were used During the first year of implementation the Chemorganisers were well received by students Increased classroom participation and confidence was noted along with stable levels of attendance Students requested the inclusion of more challenging content This intervention did not affect the time allocated for the tutorials and was easy to implement Students performed better on average than their counterparts a highly statistically significant difference was noted p

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