

3. Refer to the data table in Question 2.

a. How do the reduced ratios in the last column compare to the coefficients in the reaction shown in Model 1?

The ratios are the same as the coefficients in Model 1: 1:3:2.

b. Use mathematical concepts to explain how your answer in part a is possible.

The coefficients were all multiplied by the same number, so they all ended back to the same thing.

4. Even 538 is a small number of molecules to use in a reaction. Typically chemists use much larger numbers of molecules. (Recall that one mole is equal to 6.02×10^{23} particles.) Consider each situation below as it relates to the reaction in Model 1: $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$.

a. Calculate the amount of reactants consumed and products made.

b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole number possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$
If the reaction occurred 6.02×10^{23} times, how many molecules would be consumed or produced?	6.02×10^{23}	1.81×10^{24}	1.20×10^{24}	1 : 3 : 2
How many <i>moles</i> of each substance would be consumed or produced in the previous situation?	1	3	2	1 : 3 : 2

5. Refer to the data table in Question 4.

a. How do the reduced ratios in the last column compare to the coefficients in the reaction in Model 1?

The reduced ratios are the same as the coefficients.

b. Use mathematical concepts to explain how your answer in part a is possible.

We are still multiplying coefficients by the same number so we just didn't everything by the number to get back to a reduced ratio.

6. The ratio obtained from the coefficients in a balanced chemical equation is called the **mole ratio**.

a. What is the mole ratio for the reaction in Model 1?



b. Explain why this ratio is called the mole ratio?

It is the reduced ratio in moles (all the coefficients = 6.02×10^{23}).

Pogil Mole Ratio

Shasha Hu



Pogil Mole Ratio:

The book delves into Pogil Mole Ratio. Pogil Mole Ratio is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Pogil Mole Ratio, encompassing both the fundamentals and more intricate discussions.

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3. In chapter 2, this book will delve into the foundational concepts of Pogil Mole Ratio. This chapter will elucidate the essential principles that must be understood to grasp Pogil Mole Ratio in its entirety.
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5. In chapter 4, this book will scrutinize the relevance of Pogil Mole Ratio in specific contexts. This chapter will explore how Pogil Mole Ratio is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Pogil Mole Ratio. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Pogil Mole Ratio.

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