

Molarity

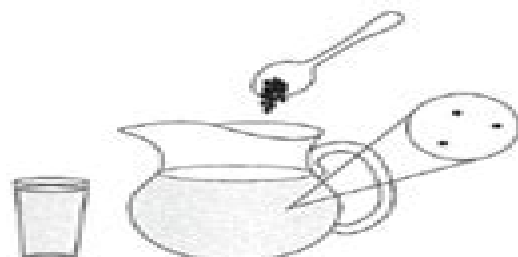
How can the concentration of a solution be expressed quantitatively?

Why?

When you buy a bottle of a certain brand of lemonade you expect it to taste just as sweet as the last time you bought that kind of lemonade. Likewise, when doctors prescribe a certain ointment, they expect the concentration of medicine to be consistent. How do companies ensure their products taste or perform the same every time you purchase them? Many companies, including pharmaceutical companies, keep track of the concentration of a solution by measuring its **molarity**—a ratio of number of solute particles to the volume of solution. In this activity you will learn about molarity and how to represent concentration quantitatively.

Model 1 – Lemonade Mixtures*

Lemonade Solution 1



Lemonade Solution 2



*Both pitchers were filled with enough water (solvent) to provide 2 liters of solution.

Dissolved Lemonade Mix particle (solute) = •

1. Refer to Model 1.
 - a. What is the solvent in this scenario?
Water.
 - b. What is the solute in this scenario?
Lemonade Mix.
 - c. What is a dissolved lemonade mix particle represented by?
A black dot.
2. Circle the word that best completes each sentence below and justify your answer based on the diagrams in Model 1.
 - a. Lemonade Solution 1 has (more/less/the same) volume of solution as 2. How do you know?
Each pitcher is filled to 2 liters.

Pogil Molarity Answers

H Kauffman

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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