

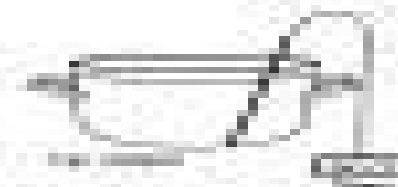
Calorimetry

What is calorimetry? How does it relate to enthalpy and temperature?

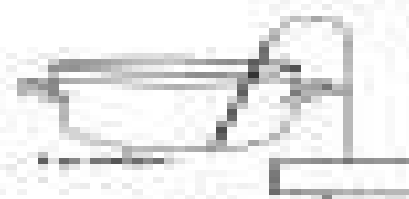
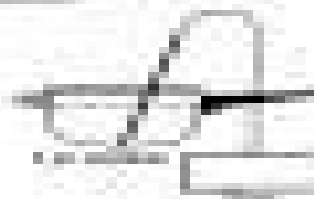
When a substance is heated, the temperature of that substance increases. Will the temperature of every other different substance be heated identical temperature increase? Will the same amount of energy be needed to raise identical temperatures for every substance? Will the same amount of the same substance be heated by the same amount of energy? How much, temperature, heat energy, and the type of substance are related.

Model 1 – A Plot of Mass

Temperature



How much energy is needed to raise 100g of water per second?



1. In Model 1, which substance had the same amount of heat? 100g

2. Compare the graphs/curves in Model 1.

a. How much total amount of energy was added to the substance? 100 J/second

b. How much total amount of energy was added to the substance? 200 J

3. In Model 1, which substance had the highest amount of energy? (The higher the temperature, the more energy.) Explain your reasoning.

Why? (What does the temperature difference tell us about the amount of energy added to the substance? What happens to the temperature of the substance when it is heated? Explain your reasoning.)

Pogil High School Chemistry Calorimetry Answer Key

Ying-Ying Zheng



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