Equilibrium

At what point is a reversible reaction "completed"?

Why?

Most of the reactions that we have studied this year have been "forward" reactions—once the reactant has changed into the product it stays that way. We can assume that eventually the reaction will "finish" when the limiting reactant runs out. However, in a reversible reaction (one that can take place in both directions), once the product is formed it can turn back into the reactant and a continuous cycle occurs. When do reactions such as these "finish?" Is there ever an end, and can we predict how much reactant or product will be present when the reaction is "done?"

Model 1 - A Reversible Reaction

$$A(g) \rightleftharpoons B(g)$$

- What is the reactant in the reaction in Model 1?
- 2. What is the product in the reaction in Model 1?
- 3. What is the significance of the double arrow in the equation in Model 1?



- Imagine that the reaction in Model 1 starts with 100 molecules of A and zero molecules of B. Explain why the concentration of substance A will never reach zero.
 - 5. Imagine the reaction in Model 1 starts with zero molecules of A and 100 molecules of B. Will the concentration of substance B increase or decrease as the reaction proceeds? Explain.

Pogil Equilibrium Key

AW Chickering



The Enigmatic Realm of Pogil Equilibrium Key: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Pogil Equilibrium Key** a literary masterpiece penned by a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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