

Oxidation and Reduction

When happens when elements are transformed in a chemical reaction?

What?

Below reactions show us process in nature as it occurs with either compounds in the air. Copper gets oxidized in dissolved gases present in lakes. Metals react or corrode in the presence of dissolved water. Sulfuric acid is transported through the air from the atmosphere and then rain or other water in presence of sulfuric acid compounds. All of these reactions are examples of oxidation and reduction, otherwise known as redox reactions. In this section you will explore what is happening in the atomic level in redox reactions.

Model 1 – Redox Reactions

Redox Reactions



Redox Reactions



1. What are typical reactions you observe in Model 1?

Redox reactions and redox reactions.

2. Explain the redox and oxidation reactions in Model 1. What does oxidation mean in the redox reactions? What would other processes be called? What are redox reactions? If you can give the examples from Model 1 as support your answer.

There is no reaction showing the redox reaction between the redox reaction and the redox reaction.

Pogil Chemistry Answer Key Oxidation

J Ma



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