

Oxidation and Reduction

When copper(II) nitrate decomposes, it is transformed to a chemical reaction:

Model 1

Below are chemical equations for reactions that involve copper(II) nitrate as the reactant. Copper gets oxidized in chemical equations involving oxygen gas. Metals react or combine in the presence of an acid solution. Nitric acid (HNO₃) is commonly found in solution for chemical reactions and can act as an oxidizing agent. In acidic solution, HNO₃ is often represented as HNO₃ and H⁺. Below are chemical equations for reactions that involve copper(II) nitrate as the reactant. In acidic solution, you will replace HNO₃ with H⁺ and H₂O in the chemical equations.

Model 1 – Redox Reactions

Redox Reactions



Redox Reactions



1. What are typical reactions for copper in Model 1?

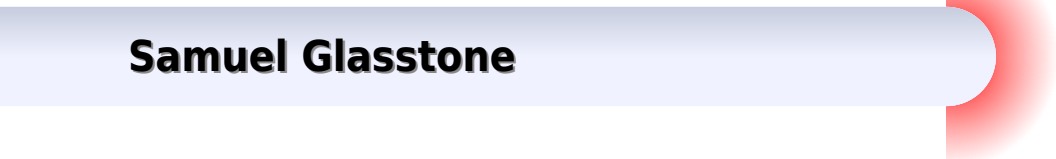
Redox reactions and acid-base reactions.

2. Examples of redox and acid-base reactions in Model 1. What does copper(II) nitrate do in the redox reactions? What does it do in the acid-base reactions? It gets oxidized to copper(II) ions. In the acid-base reactions, it acts as an oxidizing agent.

There is no reaction between copper(II) nitrate and water.

Pogil Answer Key Oxidation And Reduction

Samuel Glasstone



POGIL Answer Key Oxidation And Reduction:

Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more Empowering Science and Mathematics for Global Competitiveness Yuli Rahmawati, Peter Charles Taylor, 2019-06-07 This conference proceedings focuses on enabling science and mathematics practitioners and citizens to respond to the pressing challenges of global competitiveness and sustainable development by transforming research and teaching of science and mathematics The proceedings consist of 82 papers presented at the Science and Mathematics International Conference SMIC 2018 organised by the Faculty of Mathematics and Natural Sciences Universitas Negeri Jakarta Indonesia The proceedings are organised in four parts Science Science Education Mathematics and Mathematics Education The papers contribute to our understanding of important contemporary issues in science especially nanotechnology materials and environmental science science education in particular environmental sustainability STEM and STEAM education 21st century skills technology education and green chemistry and mathematics and its application in statistics computer science and mathematics education Chemistry Education in the ICT Age Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami, 2009-07-21 th th The 20 International Conference on Chemical Education 20 ICCE which had rd th Chemistry in the ICT Age as the theme was held from 3 to 8 August 2008 at Le M ridien Hotel Pointe aux Piments in Mauritius With more than 200 participants from 40 countries the conference featured 140 oral and 50 poster presentations th Participants of the 20 ICCE were invited to submit full papers and the latter were subjected to peer review The selected accepted papers are collected in this book of proceedings This book of proceedings encloses 39 presentations covering topics ranging from fundamental to applied chemistry such as Arts and Chemistry Education Biochemistry and Biotechnology Chemical Education for Development Chemistry at Secondary Level Chemistry at Tertiary Level Chemistry Teacher Education Chemistry and Society Chemistry Olympiad Context Oriented Chemistry ICT and Chemistry Education Green Chemistry Micro Scale Chemistry Modern Technologies in Chemistry Education Network for Chemistry and Chemical Engineering Education Public Understanding of Chemistry Research in Chemistry Education and Science Education at Elementary Level We would like to thank those who submitted the full papers and the reviewers for their timely help in assessing the papers for publication th We would also like to pay a special tribute to all the sponsors of the 20 ICCE and in particular the Tertiary Education Commission <http://tec.intnet.mu> and the Organisation for the Prohibition of Chemical Weapons <http://www.opcw.org> for kindly agreeing to fund the publication of these proceedings

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