

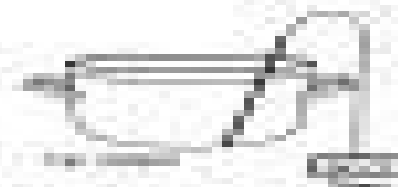
Calorimetry

What is calorimetry? How does it relate to enthalpy and temperature?

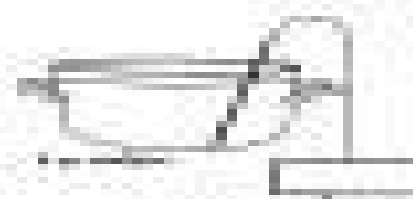
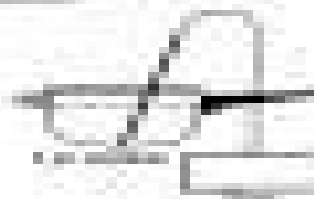
When a substance is heated, the temperature of that substance increases. Will the temperature of every other different substance be heated identical temperature increase? Will the same amount of energy be needed to raise identical temperatures for every substance at the same substance? In this practice you will explore how mass, temperature, heat energy, and the type of substance are related.

Model 1 – A Plot of Mass

Calorimetry



How much heat energy is transferred per second?



1. In Model 1, which substance had the most grams of water? 100g

2. Compare the graphs/curves in Model 1.

a. How much total joules of energy were added to the substance? 100 J/g

b. How much total joules of energy were added to the substance? 200 J

c. In Model 1, which substance had the highest mass? Which substance had the lowest mass? Explain your reasoning.

Why is it that the same amount of energy (100 J) is added to 100g of water and 200g of water? Explain how this relates to the amount of energy added to the substance.

Pogil Activities For High School Chemistry Calorimetry Answer Key

J Elliott



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