

Vapour Pressure Curves

DEFINITION

The vapour pressure of a substance depends on the temperature (Boyle's temperature leads to Boyle's vapour pressure). A liquid boils when the vapour pressure equals the atmospheric pressure. Water boils at 100°C at sea level but the pressure changes with altitude so at higher altitudes water boils at lower temperatures. Water at sea level boils at a different temperature. Various substances boil at different points (the boiling point) because they have high vapour pressure.

Learning Objectives

1. Understand the relationship between vapour pressure and temperature.
2. Understand the relationship between vapour pressure and boiling point.
3. Understand the relationship between the boiling point and the temperature of the liquid.

Boiling Point Curve

1. Understand the relationship between the boiling point and the temperature.
2. Understand the relationship between the boiling point and the temperature.
3. Understand the relationship between the boiling point and the temperature.

Properties

1. Understand the relationship between the boiling point and the temperature.

Properties

1. Understand the relationship between the boiling point and the temperature.
2. Understand the relationship between the boiling point and the temperature.
3. Understand the relationship between the boiling point and the temperature.

Properties and Relationship

Understand the relationship between the boiling point and the temperature.

1. Understand the relationship between the boiling point and the temperature.
2. Understand the relationship between the boiling point and the temperature.
3. Understand the relationship between the boiling point and the temperature.
4. Understand the relationship between the boiling point and the temperature.
5. Understand the relationship between the boiling point and the temperature.

Pogil Vapor Pressure Curves Answer Key

Yi-Tong Ma

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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