

14. Provide the chemical formula for each of the following ionic compounds.

Barium chloride $\text{Ba}^{+2} \text{Cl}^{-1} \rightarrow \text{BaCl}_2$ Magnesium oxide $\text{Mg}^{+2} \text{O}^{-2} \rightarrow \text{MgO}$

15. Consider the two chemical formulas you wrote in Question 3 for compounds of iron and sulfur. Would the name "iron sulfide" be sufficient to uniquely identify either of those compounds? Explain.

"iron sulfide" is not sufficient because there are two different forms of iron (Fe^{+2} and Fe^{+3}). The name does not indicate which iron is in the compound.

Read This!

When the metal in an ionic compound always forms an ion with the same charge, you need not indicate that charge as part of the compound name. However, some atoms have the ability to form more than one type of ion. This can make naming confusing. You can't simply refer to a compound of copper and oxygen as "copper oxide." People won't know which compound you are referring to— CuO or Cu_2O .

Model 3 – Ionic Compound Names (Metals that form multiple ions)

Cu_2O	Copper(I) oxide	PbO	Lead(II) oxide
CuO	Copper(II) oxide	PbO_2	Lead(IV) oxide
SnF_2	Tin(II) fluoride	FeCl_2	Iron(II) chloride
SnF_4	Tin(IV) fluoride	FeCl_3	Iron(III) chloride

16. Model 3 is labeled "Metals that form multiple ions." What other metals that form multiple ions could be included in Model 3? Model 1 may be helpful in this regard.

Mercury (Hg_2^{+2} and Hg^{+2}) and nickel (Ni^{+2} and Ni^{+3})

17. Describe the most obvious difference between the names in Model 3 and those in Model 2.

Roman numerals are used in Model 3 and not in Model 2.

18. Do the Roman numerals in the names in Model 3 relate to the number of cations or number of anions in the formula unit? Support your answer by citing two specific examples.

The Roman numerals have no connection between the number of cations or anions in the compound. An example is Lead (II) oxide. There is only one lead (Pb) ion in the compound and not four.

19. Keeping in mind that the sum of the charges in an ionic compound must equal zero, use the chemical formulas in Model 3 to answer the following questions:

- a. Identify the charge on the copper cations in copper(I) oxide and copper(II) oxide, respectively.

$\text{Cu}_2^+ \text{O}^{-2}$ (Each copper I must be 1^+) || $\text{Cu}^{+2} \text{O}^{-2}$ (Each copper II must be 2^+)

- b. Identify the charge on the iron cations in iron(II) chloride and iron(III) chloride, respectively.

$\text{Fe}^{+2} \text{Cl}_2^{-1}$ (Fe^{+2}) || $\text{Fe}^{+3} \text{Cl}_3^{-1}$ (Fe^{+3})

20. What do the Roman numerals in the compounds described in Question 19 indicate?

The Roman numerals indicate the charge of the ion.
(II $\rightarrow 2^+$, III $\rightarrow 3^+$)

Pogil The Mole Concept Worksheet Answers

Arturo Cuomo



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1. Understanding the eBook Pogil The Mole Concept Worksheet Answers
 - The Rise of Digital Reading Pogil The Mole Concept Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Pogil The Mole Concept Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pogil The Mole Concept Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil The Mole Concept Worksheet Answers
 - Personalized Recommendations
 - Pogil The Mole Concept Worksheet Answers User Reviews and Ratings

- Pogil The Mole Concept Worksheet Answers and Bestseller Lists
- 5. Accessing Pogil The Mole Concept Worksheet Answers Free and Paid eBooks
 - Pogil The Mole Concept Worksheet Answers Public Domain eBooks
 - Pogil The Mole Concept Worksheet Answers eBook Subscription Services
 - Pogil The Mole Concept Worksheet Answers Budget-Friendly Options
- 6. Navigating Pogil The Mole Concept Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Pogil The Mole Concept Worksheet Answers Compatibility with Devices
 - Pogil The Mole Concept Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pogil The Mole Concept Worksheet Answers
 - Highlighting and Note-Taking Pogil The Mole Concept Worksheet Answers
 - Interactive Elements Pogil The Mole Concept Worksheet Answers
- 8. Staying Engaged with Pogil The Mole Concept Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pogil The Mole Concept Worksheet Answers
- 9. Balancing eBooks and Physical Books Pogil The Mole Concept Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pogil The Mole Concept Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pogil The Mole Concept Worksheet Answers
 - Setting Reading Goals Pogil The Mole Concept Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pogil The Mole Concept Worksheet Answers
 - Fact-Checking eBook Content of Pogil The Mole Concept Worksheet Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web le bouddhisme est né des expériences et réflexions de siddhartha gautama un prince indien du VI^e siècle av J.C. après avoir mené la vie agréable de l'aristocratie indienne à 29 ans il change d'attitude et fait une série d'expériences qui devraient le mener au nirvana

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web feb 6 2019 biographie courte de bouddha gautama siddhartha connu sous le nom de bouddha ou l'Éveillé est un chef spirituel et le fondateur du bouddhisme il naît au VI^e siècle av J.C. dans la région du teraï au sud du Népal

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web la vie du bouddha cet article constitue un bref historique du bouddhisme le mot bouddha est un titre et pas un nom en lui-même il signifie celui qui est éveillé dans le sens de trouver la réalité il a d'abord été donné à un homme qui est né sous le nom de siddhartha gautama à lumbini il y a environ 2500 ans

siddhartha gautama encyclopédie de l'histoire du monde - Feb 09 2023

web sep 23 2020 siddhartha gautama plus connu comme le bouddha vers 563-483 aec était selon la légende un prince hindou qui renonça à sa position et à sa richesse pour rechercher l'illumination comme un ascète spirituel qui atteignit son but et qui en prêchant son chemin aux autres fonda le bouddhisme en Inde aux 6^e et 5^e siècles aec