

1) State the four types of chemical reactions in today's experiment.

2) Complete the table by filling in the formula for the ionic compound formed by each pair of cations and anions, as shown for the first pair.

Ion	Na^+	Ca^{2+}	Zn^{2+}	Al^{3+}
OH^-	NaOH	Ca(OH)_2	Zn(OH)_2	Al(OH)_3
Cl^-	NaCl	CaCl_2	ZnCl_2	AlCl_3
SO_4^{2-}	Na_2SO_4	CaSO_4	ZnSO_4	$\text{Al}_2(\text{SO}_4)_3$
PO_4^{3-}	Na_3PO_4	$\text{Ca}_3(\text{PO}_4)_2$	$\text{Zn}_3(\text{PO}_4)_2$	AlPO_4

3) Complete and balance the following reactions:



4) Use the solubility table below for questions a and b.

Soluble Compounds	Exceptions
Compounds containing alkali metals ions and the ammonium ion	
Nitrates, bicarbonates, and chlorates	
Halides	Halides of Ag^+ , Hg_2^{2+} , and Pb^{2+}
Sulfates	Sulfates of Ag^+ , Ca^{2+} , Sr^{2+} , Ba^{2+} , Hg_2^{2+} , and Pb^{2+}
Insoluble Compounds	Exceptions
Carbonates, phosphates, chromates, and sulfides	Compounds containing alkali metals ions and the ammonium ion
Hydroxides	Compounds containing alkali metals ions and the ammonium ion

a) Select the ionic compound which is insoluble in water.

A) NaHCO_3 B) $\text{Cu(NO}_3)_2$ C) AgBr D) MgCl_2 E) NH_4Br

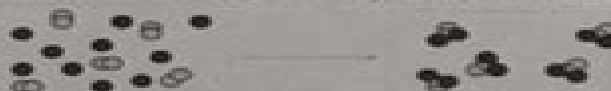
Answer: ()

b) What is the precipitate that forms when solutions of Na_3PO_4 and $\text{Fe(NO}_3)_3$ are mixed?

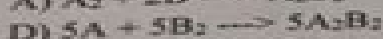
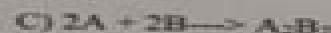
A) NaNO_3 B) NaFe C) FePO_4 D) PO_4NO_3 E) Fe_3PO_4

Answer: ()

5) The reaction of A (dark color spheres) with B (light color spheres) is shown in the following diagram:



Which equation best describes the stoichiometry of the reaction?



Answer: ()

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L Cohen



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