

4. Complete the following table for the two experiments in Model 1.

Temp	Experiment A	Experiment B
Independent Variable	# of particles	Temperature
Dependent Variable	Internal Pressure	Internal Pressure
Controlled Variable(s)	External Pressure / Volume / Temperature	External Pressure / Volume / # of particles

5. Of the variables that were controlled in both Experiment A and Experiment B in Model 1, one requires a nonflexible container. Name this variable, and explain why a nonflexible container is necessary. In your answer, consider the external and internal pressure data given in Model 1.

Experiment A Requires a nonflexible container as the independent variable is # of particles which would normally expand a container. The second experiment does not require a nonflexible container as the # of particles does not change.

 ↑ temp ≠ ↑ Volume

Read This!

Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change *how often* or *how hard* they hit. A nonflexible container is needed if the gas sample is going to have an internal pressure that is different from the external pressure. If a flexible container is used, the internal pressure and external pressure will always be the same because they are both

Pogil Gas Variables Non Flexible Containers

VM Jensen

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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