

21. Fill in the table below using what you've learned from Model 3.

Compound	Charge on Cation	Name of the Compound
PbCl_4	Pb^{4+}	Lead(IV) chloride
Fe_2O_3	Fe^{3+}	Iron(III) oxide
SnO	Sn^{2+}	Tin(II) oxide
CuBr_2	Cu^{2+}	Copper(II) bromide

22. For each of the compounds in the table below, determine the type of metal in the compound and then name the compound using the correct naming method.

	Metal forms only one ion	Metal forms multiple ions	Name
CaBr_2	X		Calcium Bromide
MgO	X		Magnesium Oxide
Ag_3N	X		Silver Nitride
SnCl_2		X	Tin(II) Chloride
CuF_2		X	Copper(II) Fluoride
K_3P	X		Potassium Phosphide
Zn_3N_2	X		Zinc Nitride
HgO		X	Mercury(II) Oxide

Pogil Activities High School Chemistry Ions Key

Guillaume Favre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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