

THE PERIODIC TABLE OF ELEMENTS

TEACHER ANSWER KEY *vocabulary practice*

WORDS:

Period/row
reactive

group/family
nonreactive

metal
conduct

nonmetal
atomic mass

metalloid

PART I: MAGIC SQUARE

Directions: Match the number of the statement best describing the word in the box and write it on the line below the word. IF they are all right, each column and row will total the same MAGIC NUMBER. Good luck!

reactive • <u>8</u>	metal • <u>3</u>	metalloid • <u>6</u>	<u>19</u>
atomic mass • <u>1</u>	group/family • <u>5</u>	nonmetal • <u>7</u>	<u>19</u>
conduct • <u>4</u>	nonreactive • <u>9</u>	period/row • <u>2</u>	<u>19</u>

TOTAL: 19 19 19

- Typically shiny, hard, solid elements that easily conduct heat and electricity. The majority of elements on the periodic table are this type of element.
- There are 7 of these on the PTE. Here you can see patterns such as increasing atomic number from left to right and decreasing reactivity of elements from left to right.
- On an element square, this number tells the total number of protons + neutrons.
- In science with relation to the periodic table, this is the ability to generate or create heat and/or electricity.
- Elements in this share similar properties, meaning they react and behave similarly to other elements found in the same column.
- An element that looks like a metal, but is a poor conductor of electricity.
- Elements that do not contain or resemble characteristics of metal. They are poor conductors of electricity.
- An element that easily and readily responds (joining, forming new substances) to other elements.
- An element that does not respond noticeably to the presence of other elements.

OVER →

Periodic Table Vocabulary Answer Key

DJ Losen



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Table of Contents Periodic Table Vocabulary Answer Key

1. Understanding the eBook Periodic Table Vocabulary Answer Key
 - The Rise of Digital Reading Periodic Table Vocabulary Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Vocabulary Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Vocabulary Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Vocabulary Answer Key
 - Personalized Recommendations
 - Periodic Table Vocabulary Answer Key User Reviews and Ratings
 - Periodic Table Vocabulary Answer Key and Bestseller Lists

5. Accessing Periodic Table Vocabulary Answer Key Free and Paid eBooks
 - Periodic Table Vocabulary Answer Key Public Domain eBooks
 - Periodic Table Vocabulary Answer Key eBook Subscription Services
 - Periodic Table Vocabulary Answer Key Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Periodic Table Vocabulary Answer Key Compatibility with Devices
 - Periodic Table Vocabulary Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Vocabulary Answer Key
 - Highlighting and Note-Taking Periodic Table Vocabulary Answer Key
 - Interactive Elements Periodic Table Vocabulary Answer Key
8. Staying Engaged with Periodic Table Vocabulary Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Vocabulary Answer Key
9. Balancing eBooks and Physical Books Periodic Table Vocabulary Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Vocabulary Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Vocabulary Answer Key
 - Setting Reading Goals Periodic Table Vocabulary Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Vocabulary Answer Key
 - Fact-Checking eBook Content of Periodic Table Vocabulary Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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