

Name: _____
Hour: _____ Date: _____

Chemistry: *The Periodic Table*

Directions: Fill in the blanks on the right with the information in the chart below.

Word List

actinide series	metal
alkali metal	metalloid
alkaline earth metal	Moseley
atomic mass	noble gas
atomic number	nonmetal
family	period
group	periodic law
halogen	periodic table
lanthanide series	transition element

Dmitri Mendeleev developed a chartlike arrangement of the elements called the __ (1) __. He stated that if the elements were listed in order of increasing __ (2) __, their properties repeated in a regular manner. He called this the __ (3) __ of the elements. The arrangement used today, devised by __ (4) __, differs from that of Mendeleev in that the elements are arranged in order of increasing __ (5) __. Each horizontal row of elements is called a(n) __ (6) __. Each vertical column is called a(n) __ (7) __, or, because of the resemblance between elements in the same column, a(n) __ (8) __.

In rows 4 through 7, there is a wide central section containing elements, each of which is called a(n) __ (9) __. Rows 6 and 7 also contain two other sets of elements that are listed below the main chart. These are called the __ (10) __ and the __ (11) __, respectively. Each of these elements, as well as those in the first two columns at the left end of the chart, is classified as a(n) __ (12) __. Each of the elements at the right side of the chart is classified as a(n) __ (13) __. Each of the elements between these two main types of elements, having some properties in common with each, is called a(n) __ (14) __.

Each of the elements in the column labeled 1 is called a(n) __ (15) __. Each of the elements in the column labeled 2 is called a(n) __ (16) __. Each of the elements in column 17 is called a(n) __ (17) __. Each of the elements in column 18 is called a(n) __ (18) __.

Periodic Table Paragraph

K Morrison

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

Periodic Table Paragraph:

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Table of Contents Periodic Table Paragraph

1. Understanding the eBook Periodic Table Paragraph
 - The Rise of Digital Reading Periodic Table Paragraph
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Paragraph
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Paragraph
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Paragraph
 - Personalized Recommendations

- Periodic Table Paragraph User Reviews and Ratings
- Periodic Table Paragraph and Bestseller Lists
- 5. Accessing Periodic Table Paragraph Free and Paid eBooks
 - Periodic Table Paragraph Public Domain eBooks
 - Periodic Table Paragraph eBook Subscription Services
 - Periodic Table Paragraph Budget-Friendly Options
- 6. Navigating Periodic Table Paragraph eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Paragraph Compatibility with Devices
 - Periodic Table Paragraph Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Paragraph
 - Highlighting and Note-Taking Periodic Table Paragraph
 - Interactive Elements Periodic Table Paragraph
- 8. Staying Engaged with Periodic Table Paragraph
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Paragraph
- 9. Balancing eBooks and Physical Books Periodic Table Paragraph
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Paragraph
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Paragraph
 - Setting Reading Goals Periodic Table Paragraph
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Paragraph
 - Fact-Checking eBook Content of Periodic Table Paragraph

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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