

## CHAPTER 6

## STUDY GUIDE

## The Periodic Table and Periodic Law

## Section 6.1 Development of the Modern Periodic Table

In your textbook, reads about the history of the periodic table's development.

Use each of the terms below just once to complete the passage.

|                     |                         |                             |                  |
|---------------------|-------------------------|-----------------------------|------------------|
| octaves             | <del>atomic mass</del>  | <del>atomic number</del>    | <del>size</del>  |
| <del>elements</del> | <del>properties</del>   | <del>Henry Mosely</del>     | <del>eight</del> |
| protons             | <del>periodic law</del> | <del>Dmitri Mendeleev</del> | <u>accepted</u>  |

The table below was developed by John Newlands and is based on a relationship called the law of (1) properties. According to this law, the properties of the elements repeated every (2) eight elements. Thus, for example, element two and element (3) nine have similar properties. The law of octaves did not work for all the known elements and was not generally (4) accepted.

|   |    |    |    |    |    |    |
|---|----|----|----|----|----|----|
| 1 | 2  | 3  | 4  | 5  | 6  | 7  |
| H | Li | G  | Be | C  | N  | O  |
| 8 | 9  | 10 | 11 | 12 | 13 | 15 |
| F | Na | Mg | Al | Si | P  | S  |

The first periodic table is mostly credited to (5) Mendeleev. In his table, the elements were arranged according to increasing (6) atomic mass. One important result of this table was that the existence and properties of undiscovered (7) elements could be predicted.

The element in the modern periodic table are arranged according to increasing (8) atomic number, as a result of the work of (9) Mosely. This arrangement is based on number of (10) protons in the nucleus of an atom of the element. The modern form of the periodic table results in the (11) periodic law which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical (12) properties.

# Periodic Table And Periodic Law Solution Manual

**C. A. Trapp, Peter Atkins, Julio dePaula**



## **Periodic Table And Periodic Law Solution Manual:**

## Decoding **Periodic Table And Periodic Law Solution Manual**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Periodic Table And Periodic Law Solution Manual**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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