OXIDATION NUMBER METHOD

Example #3: In acidic solution

- Assign oxidation numbers to each element.
- Identify the increase and decrease in oxidation numbers.
- Determine the <u>smallest whole-number ratio</u> of the oxidized and reduced elements so that the total increase in oxidation numbers equals the total decrease in oxidation numbers.
- 4) Use the determined ratio from step 3 to balance the oxidized and reduced reactants.
- Balance the other elements/compounds by inspection.
- 6. Add H,O, H*, and OH as needed to balance reactions in acidic or basic solution (recall the half-reaction method rules)

$$Cu_{(s)} + 2 NO_{3(aq)} \rightarrow NO_{2(g)} + Cu^{2+}_{(aq)}$$

Ratio = 1:2

Each Cu atom gains 2, .: 1 Cu gains 2 Each N atom loses 1, .: 2 N loses 2

Oxidation Number Yahoo Answers

Daniela Niemeyer

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Oxidation Number Yahoo Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Oxidation Number Yahoo Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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