

Oxidation - Reduction Reactions

Redox Terminology—REVIEW CHAPTER 4!!!

redox reaction -- electron transfer process



Overall process involves two Half Reactions:

oxidation -- loss of electron(s)

reduction -- gain of electron(s)



Related terms:

oxidizing agent = the substance that gets reduced (Cl_2)

reducing agent = the substance that gets oxidized (Na)

Oxidation and reduction always occur together so that there is no *net* loss or gain of electrons overall.

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Oxidation Reduction Guide Answers

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