

# ZIMSEC Nov 2023 Maths P2



Revise Maths with Eve

**ZIMBABWE SCHOOL EXAMINATIONS COUNCIL**  
General Certificate of Education Ordinary Level

**MATHEMATICS**  
PAPER 2

**4004/2**

**NOVEMBER 2023 SESSION**    2 hours 30 minutes

Additional materials:  
Mathematical Instruments  
Mathematical Tables  
Non programmable Electronic Calculator  
Plain Paper (1 sheet)  
Graph Paper (4 sheets)  
Answer Paper

## Section A solutions

# O Level Maths Paper 2 Solutions 2014 Zimsec

**R Pring**



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web 11 use avogadro s law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22 4 l at stp v 1 n 1 v 2 n 2 0 0896 l n 22 4 l 1 mole n 0 0040 moles 12 experimental molar mass g mole of butane 0 23 g 0 0040 moles 58 g mole 13 the molecular formula of butane is c 4 h 10

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web the pressure of the gas the pressure of the gas was calculated using dalton s law of partial pressures by subtracting the pressure of the water vapor from the total pressure which in this case was the atmospheric pressure 760mm hg the pressure of the water vapor was found by using the temperature of the water

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