# Polyatomic Ions WS

Ammonium  $(NH_4)^{1+}$  Dichromate  $(Cr_2O_2)^{2-}$  Nitrite  $(NO_2)^{3-}$  Phosphate  $(PO_4)^{3-}$  Carbonate  $(CO_3)^{2-}$  Hydroxide  $(OH)^{3-}$  Sulfate  $(SO_4)^{2-}$  Chromate  $(CrO_4)^{2-}$  Nitrate  $(NO_2)^{2-}$  Sulfite  $(SO_3)^{2-}$ 

#### Write the NAME of each of the following compounds.

1.	(NHL)CI Ammonium Chloride	9. Ga(ClOs): Gallium Chlorate			
2.	ee(so₄) Beryllium Sulfate	10. сы(он) "Copper I Hydroxide			
3.	(NH4)»N Ammonium Nitride	11. (NH <sub>4</sub> ) <sub>6</sub> (PO <sub>4</sub> )Ammonium Phosphate			
4,	Mgcl <sub>2</sub> Magnesium Chloride	12. Fe(SO <sub>4</sub> ) Iron II Sulfate			
5.	NH4NODAmmonium Nitrate	13. Mg(NO <sub>3</sub> ):Magnesium Nitrite			
6.	srs(PO <sub>4</sub> ) <sub>2</sub> Strontium Phosphate	14. (NH <sub>4</sub> )NO <sub>3</sub> Ammonium Nitrite			
7.	Zn(CrO <sub>4</sub> ) <sub>2</sub> Zinc Dichromate	15. Na <sub>2</sub> (Cr <sub>2</sub> O <sub>2</sub> )Sodium Dichromate			
8.	Ka(CraOn) Potassium Dichromate	16. Na(OH) Sodium Hydroxide			

#### Write the CHEMICAL FORMULA for each of the given NAMES......cross charges

17. cal	cium carbonate	Ca <sup>3</sup> * (CO <sub>3</sub> ) <sup>3</sup> *	-		Cap(CO <sub>3</sub> ) <sub>2</sub>	= CaCO <sub>3</sub>
18. barium	nitrate Ba^+2	NO3^1		Ba(NO3)		
19. ammor	nium sulfate NH4^+	1 SO4^-2		(NH4)2 SC	)4	
20. allumin	um hydraxide <u>Al^+3 C</u>	DH^-1		AI(OH)3		
21. calcium	phosphate <u>Ca^+2</u>	PO4^-3		Ca3(PO4)	2	
22. cesium	nitrate Cs NO	2^-1		CsNO3		
23. sodium	n nitrite Na^+1	NO3^-1		NaNO3		
24. cal	cium sulfate Ca^+2	SO4^-2	-	CaSO4		
25. berylliu	em sulfate Be SO4	4^-2		BeSO4		
26. sodium	carbonate Na^+1	CO3^-2		Na2CO3		
27. magne	sium phosphateMg^+2	PO4^-3	-	Mg3(PO4)	2	
		PO4^-3		Ca3(PO4)	2	

# **Polyatomic Ions Ws**

Shasha Hu



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