

Chapter Goals 1.1

Intro. to Chemical Reactions

Name: _____
 Date: _____
 Class: _____

Electrolyte Review

1. Write the formula for the following compounds:

- a) barium phosphate: $\text{Ba}_3(\text{PO}_4)_2$ b) sodium phosphate: Na_3PO_4
 c) magnesium phosphate: $\text{Mg}_3(\text{PO}_4)_2$ d) lithium nitrate: LiNO_3

2. Write the name for the following compounds:

- a) $\text{Cu}(\text{NO}_3)_2$: copper(II) nitrate b) FeCl_3 : iron(III) chloride
 c) SnSO_4 : tin(II) sulfate d) SnF_4 : tin(IV) fluoride

Chemical Reactions: What they look like...

Here is what a chemical reaction looks like when it is written:



There are examples that you are sure to recognize when you see them:

Combustion reactions react with oxygen and make carbon dioxide and water.

There are also a couple other categories: precipitation reactions and acid-base reactions. Keep up a close watch!

Critical Thinking Questions

1. Write the following reaction, using the information you know with it:



Chlorine atoms are not being added and silver atoms are being added.

Oxidation Reduction Chemquest Answer Key

AW Rasmussen



Oxidation Reduction Chemquest Answer Key:

Oxidation Reduction Chemquest Answer Key Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Oxidation Reduction Chemquest Answer Key**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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