

Oxidation – Reduction Packet



Oxidation-reduction (redox) reactions are everywhere! They take place in your body. Oxidation reactions are often responsible for the energy that keeps your body warm. They are also responsible for the rust that you see everywhere.

Oxidation-reduction reactions are also responsible for the production of energy in cells. In cells, the oxidation of glucose (sugar) provides the energy for the cell to do work. The reduction of oxygen (O₂) to water (H₂O) is a key step in the production of energy in cells.

Oxidation-reduction reactions are also responsible for the production of energy in the environment. The oxidation of organic matter by bacteria and other organisms is a key step in the production of energy in the environment. The reduction of nitrate (NO₃⁻) to ammonia (NH₃) is a key step in the production of energy in the environment.

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Remember, the total number of electrons lost in the oxidation half-reaction must equal the total number of electrons gained in the reduction half-reaction.

The overall reaction is the sum of the oxidation and reduction half-reactions. The overall reaction is the sum of the oxidation and reduction half-reactions. The overall reaction is the sum of the oxidation and reduction half-reactions. The overall reaction is the sum of the oxidation and reduction half-reactions.

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Oxidation And Reduction Packet Key

Richard Bailey



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