

- 4 Hydrogen and chlorine are reacted together to form hydrogen chloride.



- (a) Calculate the bond enthalpy for the H–Cl bond using the information in the table below.

bond	bond enthalpy / $\text{kJ mol}^{-1}$
H–H	+436
Cl–Cl	+243

Bonds broken

436

- 243

= 679

-679

Bonds made 2(HCl)

Bonds Broken + Bonds Made = -184

-679 + 2HCl = -184

2HCl = 495

HCl = 247.5

bond enthalpy = .....  $\text{kJ mol}^{-1}$  [2]

# Ocr Chemistry Past Paper F322 June 2014

**Julia Schneider**



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