

A-Level Chemistry - Year 2

OCR Chemistry A - Paper 1 June - 2019

Question - 1

1
page from a

You should spend a maximum of 40 minutes on this section.

Write your answer to each question in the box provided.

Answer all the questions.

1. In the diagrams below, each box represents an orbital and each electron is shown as an arrow. Which diagram shows the correct arrangement of electrons in an atom of carbon?

A	wrong	
B	wrong	
C	wrong	
D	wrong	

For answer

HOOO



11

OCR

International Examinations

Tuesday 4 June 2019 – Afternoon

A Level Chemistry A

H432091 Periodic table, elements and physical chemistry

Time allowed: 2 hours 15 minutes

Useful facts:

1 mole has a mass of 12 g for carbon-12
and 16 g for oxygen-16.

Useful facts:

1 mole has a mass of 12 g for carbon-12
and 16 g for oxygen-16.



Please write clearly in black ink. Do not write in the brackets.

Centre number

Candidate number

First name

Last name

INSTRUCTIONS

- 1. Use black ink. You may use an HB pencil for graphs and diagrams.
- 2. Answer all the questions.
- 3. Where appropriate, your answers should be supported with writing. Marks may be given for a correct method even if the answer is incorrect.
- 4. Write your answer to each question in the space provided. If additional space is required, use the back page(s) at the end of this booklet. The question numbers must be clearly shown.

INFORMATION

- 1. The total mark for this paper is 100.
- 2. The marks for each question are shown in brackets ().
- 3. Quality of extended responses will be assessed in questions marked with an asterisk (*).
- 4. This document consists of 10 pages.

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