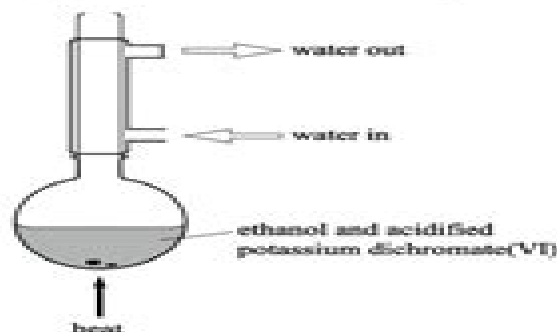


This resource includes exam questions from both Chemistry A and Chemistry B. It is not designed as a topic test but is instead intended as a document to support the teaching and learning of Module 1 – Development of practical skills in chemistry.

1. Ethanol is oxidised to ethanoic acid using acidified potassium dichromate(IV) solution. The reaction is heated under reflux using the equipment shown in the diagram below.



What is the reason for heating under reflux?

- A. to ensure even heating
- B. to prevent any substances escaping
- C. to boil the mixture at a higher temperature
- D. to allow efficient mixing

Your answer ☐

[1]

2. A chemist compares the rates of hydrolysis of 1-chloropropane and 1-bromopropane in ethanol. Which reagent in aqueous solution should be used?

- A Silver chloride
- B Silver nitrate
- C Potassium chloride
- D Potassium nitrate

Your answer ☐

[1]

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Joacim Rocklöv



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