Lesson Quiz

Observing Chemical Change

Write the letter of the correct answer on the line at the left.

- Which of the following is true about chemical reactions?
 - A They are accompanied by changes in energy.
 - B They form new substances with new properties.
 - C both A and B
 - D neither A nor B
- Which of the following is NOT a physical property?
 - A melting point
 - B state of matter
 - C density
 - D flammability

- In an endothermic reaction, energy is
 - A absorbed
 - B released
 - C converted to mass
 - D synthesized
- Substances formed as a result of a chemical reaction are called
 - A catalysts
 - B precipitates
 - C products
 - D reactants

If the statement is true, write true. If the statement is false, change the underlined word or words to make the statement true.

- In an exothermic reaction, products have more energy than reactants.
- Water boils at 100°C. This is an example of a chemical property.
- Substances that enter into a chemical reaction are called products.
- The ability to react with oxygen is an example of a <u>chemical</u> property.
- Another name for a chemical change is a chemical <u>bond</u>.
- In a physical change, some of the physical properties of the substance may be altered and the chemical composition remains the same.

Observing Chemical Changes Workbook Answers

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Observing Chemical Changes Workbook Answers:

Observing Chemical Changes Workbook Answers Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has be much more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Observing Chemical Changes Workbook Answers**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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