

Periodic Table WebQuest

Periodic Table of the Elements

Legend:

- Lanthanide Series
- Actinide Series

Directions: In each of the websites listed, find the information asked for.

Use <http://atomictimeline.net/index.php> to write the contributions of each person toward the development of their theory of the atom (atomic model) and the Periodic Table of Elements.

1. Democritus
2. John Dalton
3. J. J. Thompson
4. Niels Bohr
5. Ernest Rutherford

Use

http://www.kyrene.org/Staff/mijohns/atomic_structure_website/electroncloud.htm, http://education.jlab.org/qa/particlemass_02.html, and <http://www.colorado.edu/physics/2000/isotopes/index.html>, to answer the following:

6. List 3 unique aspects of the electron cloud model in general comparison to previous models.
7. How does the mass of a proton compare to the mass of a neutron?
8. How does the mass of a proton compare to the mass of an electron?

Periodicity Webquest Answer Key

PT Brinkman

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the PT Brinkman text.

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the order of the reaction with respect to a the value of n is not related to the reaction stoichiometry and

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web the rate determining step limits the overall rate and therefore determines the rate law for the overall reaction introduction multistep reactions one of the most important applications of kinetics is to the study of reaction mechanisms or the sequences of steps by which chemical reactions occur

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web the initial rate law is then measured for each of the reactions consider the reaction between nitrogen monoxide gas and hydrogen gas to form nitrogen gas and water vapor the following data were collected for this reaction at 1280 text O text c see table below notice that the starting concentrations of CeNO and CeH

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web determination of a rate law and activation energy in this experiment glass beakers pipettes and a hotplate were utilized

to set up the procedures a solution of deionized water a buffer potassium iodide KI sodium thiosulfate $\text{Na}_2\text{S}_2\text{O}_3$ starch and hydrogen peroxide H_2O_2 were used in order to calculate the rate of reaction

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web oct 27 2022 the rate law rate k $[\text{H}^+]$ describes a reaction that is first order in $[\text{H}^+]$ first order in $[\text{OH}^-]$ and second order overall example 12 3 1 writing rate laws from reaction orders an experiment shows that the reaction of

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web rate laws sometimes called differential rate laws or rate equations are mathematical expressions that describe the relationship between the rate of a chemical reaction and the concentration of its reactants as an example consider the reaction described by the chemical equation $\text{a} + \text{b} \rightarrow \text{products}$

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