

## Periodic Table Named-Groups Worksheet – 2

**DIRECTIONS:** Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

A diagram of a 10x10 grid representing a 2D plane. The grid is divided into four quadrants by a horizontal line and a vertical line. The quadrants are labeled A (top-left), B (top-right), C (bottom-left), and D (bottom-right). The grid contains various letters and numbers. The top row is labeled A and B. The second row contains C, D, E, F, G, H, I, J, K, L. The third row contains M, N, O, P, Q, R, S, T, U, V, W, X, Y, Z. The bottom row contains A, B, C, D, E, F, G, H, I, J, K, L, M, N, O, P, Q, R, S, T, U, V, W, X, Y, Z. Arrows point from the letters R and S in the third row to the letters V and W in the bottom row.

List elements in alphabetical order when more than 1 element answers a question.

- \_\_\_\_\_ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- \_\_\_\_\_ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- \_\_\_\_\_ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- \_\_\_\_\_ 4) Which letter labels the halogen element with the lowest atomic number?
- \_\_\_\_\_ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- \_\_\_\_\_ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

# Periodic Tables Answer Key

**Neil D. Jespersen, Pamela Kerrigan**



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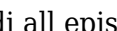

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