



Name: ANSWER KEY  
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# THE PERIODIC TABLE

## Review Worksheet

1. Fill in the blanks with the terms and words most appropriate to complete each sentence.

- a) In the mid-nineteenth century, only sixty or so elements had been discovered. In 1869 (a year) a Russian chemist and inventor named Dmitri Mendeleev created a chart called the periodic table that predicted the properties of elements that had not yet been discovered. He did this by arranging the elements according to their increasing atomic numbers and by their physical and chemical properties.
- b) Each element in the periodic table is represented by an element box. In this box there is the following information: two numbers, one is a whole number which is called the atomic number, and the other number is larger and contains a decimal, and this one is called the atomic mass. In the box, there is also the short form of the element called the element symbol, and then the full identity of the element which is written using the element name.
- c) Today the modern periodic table is arranged into columns and rows. The columns are called groups and there are 18 (a number) of them. The rows are called periods and there are 7 (a number) of them. The periodic table is divided by a zigzag line also known as the staircase that separates the metals, found on the left of it, from the non-metals, found on the right of it. Directly around the zigzag line are eight elements called metalloids.
- d) All the metals, except for mercury, which is a liquid, are found as solids at room temperature. The metals are also mostly shiny in color, are ductile, malleable, conduct electricity and some are attracted to magnets.
- e) At room temperature, most of the non-metals are found in the gas state except for bromine which is a liquid, and iodine selenium carbon, phosphorus and sulfur which are all found as solids. These elements are colorless or come in various colors, are not ductile, not malleable and do not conduct electricity nor are they attracted to magnets.
- f) Metalloids are elements that have some properties of metals and some properties of non-metals. They are all found in the solid state. Most have a silvery metallic luster, they are brittle and are medium to good electrical conductors.
- g) Chemical families are groups on the periodic table that contain elements that share similar properties.
- h) Group 1 elements are called the alkali metals and they contain the most reactive group of all the metals. They react very violently with water and therefore must be stored under mineral oil so that the water vapor in the air cannot react with these elements. These metals get more reactive as you go down (up or down) the group and the most reactive of all the elements in this group is francium.
- i) Group 2 elements are called the alkaline earth metals and they contain the second most reactive group of all the metals. They also react with water and get more reactive as you go down (up or down) the group.

# Periodic Table Review Sheet Answers

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Fryhle, Scott A. Snyder**



## **Periodic Table Review Sheet Answers:**

## Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Periodic Table Review Sheet Answers**

In a global inundated with monitors and the cacophony of instant communication, the profound energy and mental resonance of verbal beauty often diminish in to obscurity, eclipsed by the continuous assault of sound and distractions. Yet, situated within the musical pages of **Periodic Table Review Sheet Answers**, a captivating function of literary elegance that impulses with raw thoughts, lies an unforgettable trip waiting to be embarked upon. Penned by way of a virtuoso wordsmith, that exciting opus courses viewers on a psychological odyssey, lightly exposing the latent potential and profound affect stuck within the elaborate web of language. Within the heart-wrenching expanse with this evocative examination, we shall embark upon an introspective exploration of the book is main styles, dissect its fascinating publishing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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web jun 17 2023 du hast schleier vor den augen keine sorge das ist kein grund zur panik wir erklären dir in diesem artikel was du dagegen tun kannst wenn du plötzlich schleier vor den augen hast wir geben dir ein paar tipps und tricks die dir helfen können das problem schnell in den griff zu bekommen

warum sehe ich sterne vor meinen augen erfahre die ursache - May 01 2022

web may 14 2023 du siehst sterne vor deinen augen weil du unter einer kurzzeitigen durchblutungsstörung im gehirn leidest diese störung tritt auf wenn der blutfluss zu deinem kopf nicht richtig funktioniert was zu einer plötzlichen unterversorgung mit sauerstoff führt das kann zu einer art kurzen stromausfall im kopf führen der die

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web may 21 2023 hast du schonmal das gefühl gehabt dass vor deinen augen kurz alles flimmert das ist bei vielen menschen ganz normal und es gibt einige gründe warum es passiert in diesem artikel werden wir uns gemeinsam anschauen warum vor deinen augen flimmern kann