

Types of Chemical Reactions Lab Answer Key

Chemical Reaction #1

<i>Reactants (write in words)</i>	hydrochloric acid and magnesium
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	single replacement
<i>Predict Products (write in words)</i>	magnesium chloride and hydrogen gas
<i>Equation (write in symbols)</i>	$2\text{HCl}_{\text{(aq)}} + \text{Mg}_{\text{(s)}} \rightarrow \text{MgCl}_{2\text{(aq)}} + \text{H}_{2\text{(g)}}$

Chemical Reaction #2

<i>Reactants (write in words)</i>	copper metal and oxygen gas
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	synthesis
<i>Predict Products (write in words)</i>	copper oxide
<i>Equation (write in symbols)</i>	$2\text{Cu}_{\text{(s)}} + \text{O}_{2\text{(g)}} \rightarrow 2\text{CuO}_{\text{(s)}}$

Chemical Reaction #3

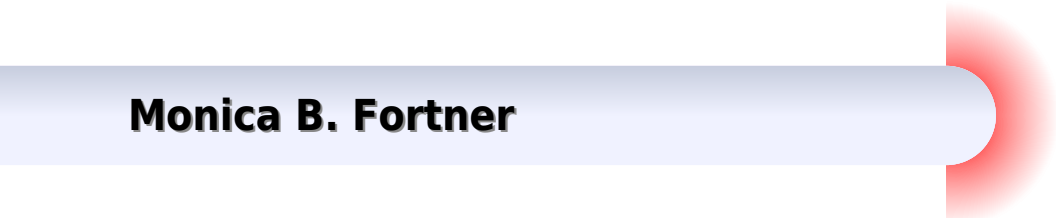
<i>Reactants (write in words)</i>	methane gas and oxygen gas
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	combustion
<i>Predict Products (write in words)</i>	carbon dioxide and water
<i>Equation (write in symbols)</i>	$\text{CH}_{4\text{(g)}} + 2\text{O}_{2\text{(g)}} \rightarrow \text{CO}_{2\text{(g)}} + 2\text{H}_2\text{O}_{\text{(g)}}$

Chemical Reaction #4

<i>Reactants (write in words)</i>	hydrogen peroxide
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	decomposition
<i>Predict Products (write in words)</i>	water and oxygen gas
<i>Equation (write in symbols)</i>	$2\text{H}_2\text{O}_{2\text{(l)}} \rightarrow 2\text{H}_2\text{O}_{\text{(l)}} + \text{O}_{2\text{(g)}}$

Novanet Answer Key Chemistry

Monica B. Fortner



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NovaNET User Manual for C-router Instructors Monica B. Fortner,1990 *The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success (Volume One) (Preliminary E* Rachel

Turoscy,2017-12-31 Modern Chemistry Holt, Rinehart and Winston Staff,2006-01 **The Answer Key: A**

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Chemistry: the Physical Setting Prentice Hall (School Division), Twelve independent units supported by numerous illustrations sample problems vocabulary and review questions offer students comprehensive preparation for the Regents Examination for The Physical Setting Chemistry Hundreds of practice questions written and organized in the format of the exam also include point values for each type of question [The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success](#) Rachel Turoscy, 2017-12-31

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